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(54) Title: BETA-AMYLOID PEPTIDE INHIBITORS

(57) Abstract: The present invention relates to compounds which inhibit the binding of metal ions to a region in the N-terminal loop of the β -amyloid peptide which includes a cluster of histidine residues. In addition, the invention relates to pharmaceutical compositions including these compounds as the active agent, and to methods of treatment involving the administration of these compounds. The compounds of the invention are useful in the treatment of Alzheimer's disease and other amyloid-related conditions. In a first aspect the present invention provides a compound which interacts with the β -amyloid peptide in such a way that the N-terminal loop of the peptide (amino acid residues 1-15) is blocked or destabilised, thereby inhibiting the binding of one or more metal ions to at least one histidine residue within the N-terminal loop. Preferably the compound inhibits binding of Cu^{2+} , Zn^{2+} and Fe^{3+} ions, but not Mg^{2+} or Ca^{2+} ions.

BETA-AMYLOID PEPTIDE INHIBITORS

The present invention relates to compounds which inhibit the binding of metal ions to the N-terminal region of the β -amyloid peptide. In addition, the present invention relates to pharmaceutical compositions including these compounds as the active agent, and to methods of treatment involving the administration of these compounds. The compounds of the invention are useful in the treatment of Alzheimer's disease.

BACKGROUND OF THE INVENTION

Alzheimer's disease (AD) is characterised by the presence of distinctive lesions in the victim's brain. These brain lesions include abnormal intracellular filaments called neurofibrillary tangles, and extracellular deposits of amyloid in senile, or amyloid, plaques. Amyloid deposits are also present in the walls of cerebral blood vessels of Alzheimer's patients.

The major constituent of amyloid plaques has been identified as a 4 kilodalton peptide (39-43 residues) called β -amyloid peptide ($A\beta$) (Glennner and Wong, 1984). Diffuse deposits of $A\beta$ peptides are frequently observed in normal adult brains, whereas Alzheimer's disease brain tissue is characterised by more compacted, dense-core β -amyloid plaques. These observations suggest that $A\beta$ deposition precedes, and contributes to, the destruction of neurons that occurs in Alzheimer's disease. In further support of a direct pathogenic role for AD, β -amyloid has been shown to be toxic to mature neurons both in culture and *in vivo* (Yanker et al., 1989).

Natural $A\beta$ is derived from proteolysis from a much longer protein known as the amyloid precursor protein (APP) (Kang, J et al, 1987). The APP gene maps to chromosome 21, thereby providing an explanation for the β -amyloid deposition seen at an early age in individuals with Down's syndrome, which is caused by trisomy of chromosome 21.

A β peptides are cleaved from APP, and then undergo aggregation to produce the insoluble toxic β -sheet structures which are found in extracellular deposits in Alzheimer's disease and Down's syndrome. Recent data suggest that the aggregated peptide has redox properties and can generate reactive oxygen species, which attack enzymes and possibly cell membranes, causing neurotoxicity (Markesbery, W.R. 1997). Antitioxidants are known to protect against A β -induced toxicity.

A β has been shown to bind copper and iron in stoichiometric amounts, with the associated formation of reactive oxygen species such as peroxides and hydroxide radicals, which are possible sources of the neurotoxicity (Bush et al., 1998). While the formation of peroxide in post-mortem samples of Alzheimer's disease brain has been observed, there was little peroxide formation in control tissue (Cherny et al., 1998). The peroxidase activity observed in the samples of Alzheimer's disease brain was abolished when treated with certain chelators (Cherny et al., 1998). The formation of reactive oxygen species was accompanied by a reduction in the valence state of the metal, ie Cu(II) to Cu(I) and Fe(III) to Fe(II) (Atwood et al., 1998a). Reactive oxygen species can also lead to free radical formation on the A β peptide, which leads over time to covalent cross-linking of the A β peptides (Bush et al., 1998). In addition, a number of metal ions, including Zn, Ni and Cu, have been shown to induce aggregation of A β (Atwood et al., 1998b). When brain tissue from both control and Alzheimer's disease-affected subjects was treated with chelators which are specific for zinc and copper, there was greatly enhanced solubilisation of A β , with an increase of up to 700%, suggesting that zinc and copper play a role in the assembly of the A β deposits (Cherny et al., 1998).

Histidine residues have been implicated in the binding of metal ions to A β peptides. For instance rat A β 1-40, in which His13 is mutated to Arg, does not aggregate, nor does A β 1-40 treated with diethyl

pyrocarbonate, which binds to the imidazole nitrogen of histidine (Atwood et al., 1998). Subsequently to the priority date of this application, it was reported that three histidine residues in the N-terminal hydrophilic region of human A β provide primary metal binding sites, and that the solubility of the complex between metal and A β depends on the mode of metal binding. The authors proposed that Cu²⁺ would protect A β against Zn-induced aggregation by competing with zinc ions for binding sites on the histidine residues (Miura et al., 2000).

In contrast, we propose that inhibition of binding of zinc, copper and/or iron to the A β peptide will have significant therapeutic value in the treatment of Alzheimer's disease.

It has been reported that certain tetrapyrroles, especially certain porphyrin and phthalocyanine compounds inhibit conversion of normal, protease-sensitive prion protein (PrP^{sen}) to the protease-resistant form (PrP^{res}) which is implicated in the pathogenesis of transmissible spongiform encephalopathies (TSEs) such as Creutzfeldt-Jacob disease (Caughey et al., 1998), and that three of these compounds inhibited TSE disease *in vivo* (Priola et al., 2000). However, both metal-free and metal-complexed tetrapyrroles were active, and the authors considered that the mechanism of action involved direct interaction between the compound and the infectious agent. Although the authors speculated that the compounds might also be useful in the treatment of non-prion mediated amyloid-related conditions, such as Alzheimer's disease or Type II diabetes, this was no more than speculation (Priola et al., 2000). Moreover, all of the compounds disclosed have multiple substitutions on the tetrapyrrole ring, whereas the tetrapyrrole compounds of the present invention are preferably substituted only on one of the rings.

It will be clearly understood that, although a number of prior art publications are referred to herein, this reference does not constitute an admission that any of

these documents forms part of the common general knowledge in the art, in Australia or in any other country.

SUMMARY OF THE INVENTION

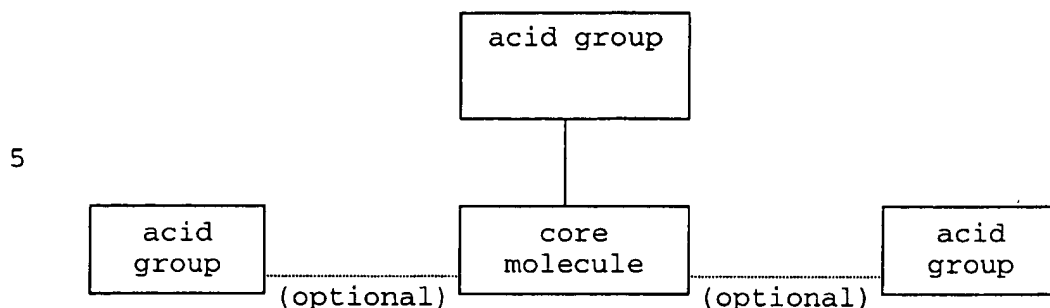
5 The present inventors have now found that zinc and copper bind predominantly to a region in the N-terminal loop of A β which includes a cluster of histidine residues. This finding provides the basis for the rational design or selection of inhibitors of the binding of zinc, copper
10 and/or iron to A β .

 Accordingly, in a first aspect the present invention provides a compound which interacts with the β -amyloid peptide in such a way that the N-terminal loop of the peptide (amino acid residues 1-15) is blocked or
15 destabilised, thereby inhibiting the binding of one or more metal ions to at least one histidine residue within the N-terminal loop.

 Preferably the compound inhibits binding of Cu²⁺, Zn²⁺ and Fe³⁺ ions, but not Mg²⁺ or Ca²⁺ ions.

20 Preferably the compound has a conformation and polarity such that it binds to at least one, more preferably at least two, and more preferably three histidine residues in the N-terminal loop, selected from the group consisting of His6, His13 and His14. More
25 preferably the compound also binds to at least one additional amino acid in the N-terminal loop, selected from the group consisting of Asp7, Tyr10, and Glu11.

 The compound may have acidic groups which interact with one or more of the His residues in the N-terminal
30 loop. For example, the compound may be represented as follows:



10 wherein the core molecule has a conformation and polarity such that the acid group(s) interact with one or more of His6, His13 and His14. The acid may be any acid group, including, but not limited to, CO_2H , PO_3H_2 , SO_3H , OSO_3H_2 , OPO_3H_2 and the like.

15 The compound may be a molecule with one to three carboxylic acid groups, the length of the molecule being such that it can be received within the N-terminal loop, and such that at least one carboxyl group is in proximity to at least one of the histidine residues. Without wishing
20 to be bound by theory, we believe that it is likely that such molecules will have a molecular mass in the region of 2000 Daltons.

The compound may be an organic molecule, a peptide or a metal complex. In this aspect of the invention, however,
25 it is preferred that the compound is not a metal complex. Preferably the compound has overall hydrophobic character. More preferably the compound is able to penetrate the blood-brain barrier.

In a particularly preferred embodiment of the
30 invention, the inhibitor compound comprises, or is conjugated to, a targeting moiety.

The term "targeting moiety" as used herein refers to a functional group which will specifically interact with the β -amyloid peptide. That is, the inhibitor compound
35 includes or is covalently linked to a targeting moiety which will specifically bind to or associate with the β -amyloid peptide. Suitable targeting moieties include,

but are not limited to, polypeptides, nucleic acids, carbohydrates, lipids, β -amyloid ligands, antibodies, dyes and the like. In a preferred embodiment the targeting moiety has a hydrophobic region which interacts with the
5 tail of the β -amyloid peptide. For example, the targeting moiety may include a fatty acid molecule.

Preferably the targeting moiety targets the compound to the site defined by residues 15-21 of the β -amyloid peptide. The targeting moiety may be a peptide which
10 comprises a sequence which corresponds to that of residues 15-21 of the β -amyloid peptide. More preferably the inhibitor-targeting moiety complex is able to penetrate the blood-brain barrier.

In a second aspect, the invention provides a method
15 of selecting or designing a compound which inhibits the binding of metal ions to the N-terminal loop of the β -amyloid peptide, which method comprises the steps of

(i) selecting or designing a compound which has a conformation and polarity such that it binds to at least
20 one, more preferably at least two and more preferably three amino acids in the N-terminal loop, selected from the group consisting of His6, His 13 and His14; and

(ii) testing the compound for the ability to inhibit binding of metal ions to the N-terminal loop of the
25 β -amyloid peptide.

Preferably the compound inhibits binding of Cu^{2+} , Zn^{2+} and Fe^{3+} ions, but not Mg^{2+} or Ca^{2+} ions.

Preferably the compound has a conformation or polarity such that it also binds to at least one amino acid
30 in the N-terminal loop, selected from the group consisting of Asp7, Tyr10, and Glu11. Preferably the compound also has overall hydrophobic character. More preferably the compound is able to penetrate the blood-brain barrier.

In a third aspect, the invention provides a compound
35 which inhibits the binding of metal ions to the N-terminal loop of the β -amyloid peptide, wherein the compound is

obtained by a method according to the second aspect of the invention.

In a fourth aspect, the invention provides a composition comprising a compound according to the first or the third aspects of the present invention, together with a pharmaceutically acceptable carrier. Methods and pharmaceutical carriers for preparation of pharmaceutical compositions are well known in the art, as set out in textbooks such as Remington's Pharmaceutical Sciences, 19th Edition, Mack Publishing Company, Easton, Pennsylvania, USA.

The compounds of the present invention may be formulated into pharmaceutical compositions, and administered in therapeutically effective doses. The term "therapeutically effective dose" means a dose which results in the inhibition of natural binding of metal ions to the N-terminal loop of the β -amyloid peptide. The pharmaceutical compositions may be administered in a number of ways, including, but not limited to, orally, subcutaneously, intravenously, intraperitoneally and intranasally. The most appropriate dose and route of administration will be dependent on the age and general state of health of the subject to be treated, and will be at the discretion of the attending physician. This dose can be readily ascertained by one skilled in the art, using well-known techniques.

In a fifth aspect, the invention provides a method of inhibiting the binding of one or more metal ions to the β -amyloid peptide, or inhibiting the aggregation of β -amyloid peptide, which method comprises the step of exposing the peptide to a compound which blocks or destabilises the N-terminal loop of the peptide, thereby inhibiting the binding of one or more metal ions to at least one histidine residue within the N-terminal loop.

Preferably the compound has a conformation and polarity such that it binds to at least one, more preferably at least two, and more preferably three

histidine residues in the N-terminal loop of the β -amyloid peptide, selected from the group consisting of His6, His13 and His14. More preferably the compound also binds to at least one additional amino acid in the N-terminal loop,
5 selected from the group consisting of Asp7, Tyr10, and Glu11.

In a particularly preferred embodiment, the compound is a metal complex which can exchange or bind functional moieties such as histidine, with the proviso that the
10 compound is not haemin or haematin. Preferably the metal complex is capable of binding between 1 and 3, preferably 2 or 3, histidine residues of the N-terminal loop of the β -amyloid peptide. The complex may bind to other residues in addition to the histidine residues. More preferably the
15 complex also binds to at least one additional amino acid in the N-terminal loop, selected from the group consisting of Asp7, Tyr10, and Glu11.

Metal ions capable of binding to the imidazole nitrogen(s) of histidine include Mn, Fe, Co, Ni, Cu, Zn,
20 Ru, Pd, Ag, Cd, Pt, Au, Rh and Hg. Complexes of these metals are expected to be predominantly four coordinate tetrahedral (distorted tetrahedral)/square planar complexes, five coordinate complexes with either a trigonal bipyramid or square pyramid configuration, or six
25 coordinate octahedral (or distorted octahedral) complexes.

Even more preferably the inhibitor compound comprises, or is conjugated to, a targeting moiety. Preferably the targeting moiety targets the compound to the site defined by residues 15-21 on the β -amyloid peptide.

30 In a further preferred embodiment, the inhibition of binding of one or more metal ions to the β -amyloid peptide occurs *in vivo*.

In a sixth aspect, the invention provides a method of prevention, treatment or alleviation of Alzheimer's disease
35 which method comprises the step of administering a compound or a pharmaceutical composition according to the invention to a subject in need of such treatment.

The patient is monitored for clinical improvement, which may commence within as little as one week, but more probably may be observed at six weeks, and may take as long as 12 months. The normal clinical indices which are used in
5 the monitoring of patients with the relevant condition are used. Where the treatment is prophylactic, the patient is monitored for signs of development of the condition. The attending clinician will be aware of the most suitable tests to use.

10 Where this method is to be used for prophylactic purposes, the subject is preferably one at increased risk of developing the condition. For example, the subject may have one or more family members with the condition, eg. familial Alzheimer's disease, or may have trisomy of
15 chromosome 21 (Down's syndrome).

It will be clearly understood that, for the purposes of this aspect of the invention, the compound is not haemin.

In a seventh aspect, the invention provides a method
20 of monitoring the efficacy of treatment according to the method of the invention, comprising the steps of obtaining a sample of a biological fluid for a patient undergoing treatment, and measuring the level of A β in the sample, in which increased A β levels compared to levels of A β in a
25 normal control sample are indicative of the efficacy of the treatment.

For the purposes of this specification it will be clearly understood that the word "comprising" means "including but not limited to", and that the word
30 "comprises" has a corresponding meaning.

BRIEF DESCRIPTION OF THE FIGURES

Figure 1 is a representation of the β -amyloid peptide showing a structured turn in the region of amino acid
35 residues 15-21.

Figure 2 shows a model of Zn bound to the three histidine residues of A β 1-40.

Figure 3 is a NMR spectrum showing the effect of Zn^{2+} binding to A β 1-28.

Figure 4 is a NMR spectrum showing the effect of Cu^{2+} binding to A β 1-28.

5 Figure 5 is a NMR spectrum showing the effect of addition of Cu^{2+} and cobalt complex to A β 1-28.

Figure 6 is a NMR spectrum showing the binding of the cobalt complex to A β 1-28.

10 Figure 7 is a Western blot showing results of brain tissue assays testing the ability of a range of metal compounds to solubilize A β deposits.

Figure 8 shows a model of the cobalt-corrin ring complex bound to A β 1-40.

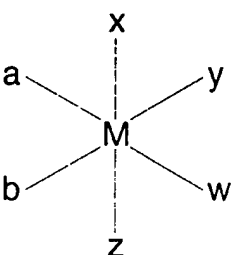
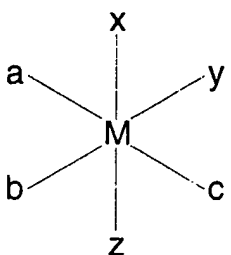
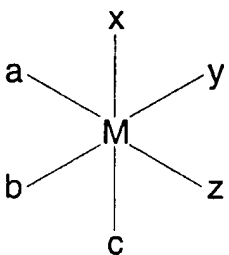
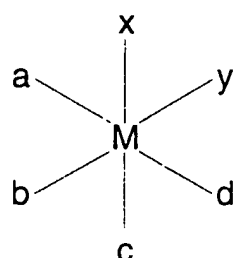
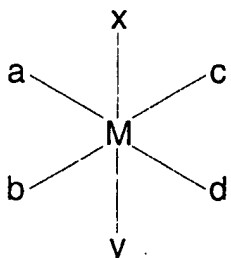
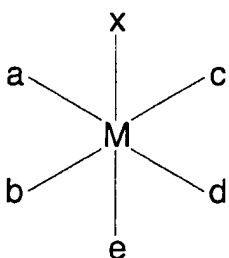
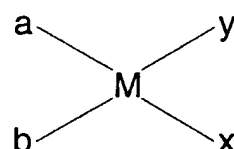
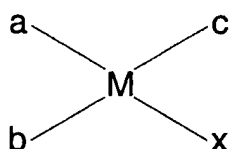
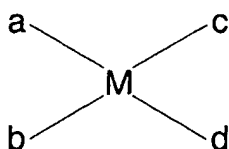
15 Figure 9 shows NMR spectra demonstrating the binding of the compound KJB001 to A β 1-28.

DETAILED DESCRIPTION OF THE INVENTION

The present inventors have developed three-dimensional structural information concerning the
20 N-terminal region of the β -amyloid peptide, and have identified a cluster of three histidine residues which constitute a binding site for metal ions. This information provides a rational basis for the development of compounds which inhibit the binding of metal ions to the N-terminal
25 loop of the β -amyloid peptide. Such inhibitors have the potential to inhibit aggregation of β -amyloid peptides and to reduce metal-induced neurotoxicity. Accordingly, these inhibitors are likely to have therapeutic value in the treatment of diseases such as Alzheimer's disease.

30 Using the structural information provided by the inventors, the general principles of drug design can be applied by persons skilled in the art to produce compounds which preferentially bind to at least one of the histidine residues in the N-terminal loop (ie His6, His13 or His14),
35 and inhibit the binding of metal ions to the N-terminal loop of the amyloid peptide.

Preferred inhibitors within the context of the present invention include metal complexes which can exchange or bind functional moieties such as histidine. Preferably the metal complex is capable of binding between
5 1 and 3, preferably 2 or 3, histidine residues of the N-terminal loop of the β -amyloid peptide. The complex may bind to other residues in addition to the histidine residues. Metal ions capable of binding to the imidazole nitrogen of histidine include Mn, Fe, Co, Ni, Cu, Zn, Ru,
10 Pd, Ag, Cd, Pt, Au, Rh and Hg . Complexes of these metals are expected to be predominantly four coordinate tetrahedral (distorted tetrahedral)/square planar complexes, or six coordinate octahedral (or distorted octahedral) complexes. In the case of the four coordinate
15 complexes they could react with the β -amyloid peptide to replace ligands; alternatively, by binding to His residues as well as to their initial ligand(s) their coordination number is increased to 5 or 6. Five coordinate complexes with either a trigonal bipyramid or square pyramid
20 configuration may also be used. Examples of suitable complexes are shown below.



In these formulae:

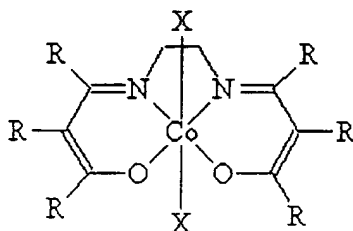
- (i) a, b, c, d and e are non-leaving groups,
 5 preferably chelating groups including Schiff bases, porphyrin rings, macrocycles, polyamino-carboxylates, heterocyclic aromatic groups such as 2,2'-bipyridine and 1,10-phenanthroline, peptides, nucleobases, or chelating ligands in which one of the donor atoms is a phosphine
 10 phosphorus atom;
- (ii) M is a metal; and
- (iii) w, x, y and z are leaving groups (ie those groups which will be replaced by histidine and possibly other residues when the metal complex reacts with the
 15 β -amyloid peptide), and include halogens, amines, ammonia, pyridyls, imidazoles, nucleobases, peptides, H₂O/OH, carboxylic acids, phosphates, sulfates, nitrate, triflate, or alkoxides.

The term " nucleobase" means a purine or a pyrimidine, or an analogue thereof. It will be appreciated that multidentate macrocyclic ligands may have a variety of donor atoms, and that it is possible that one or more of
5 the non-leaving groups could be a stable monodentate ligand such as cyanide, or an organic group such as a methyl group.

Those skilled in the art will recognise that the appropriate combination of non-leaving and leaving groups
10 will be dependent on the identity of the metal.

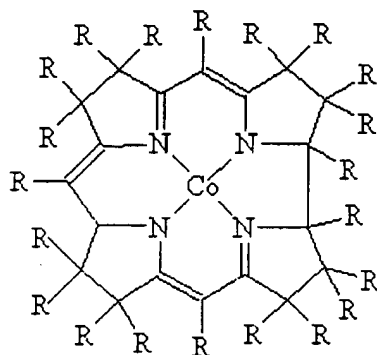
Metal complexes which have the potential to bind to histidine residues are described in WO 97/21431 and WO 96/18402, the entire contents of which are incorporated herein by this reference. Other examples of complexes which
15 may act as inhibitors are as follows:

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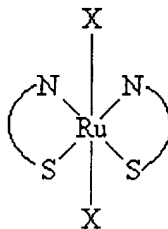
10



II

15

20



III

25

In formulae I, II and III, the X and R groups may be any suitable leaving or targeting groups. For example, X and R may be the same or different, and include, without being limited to, the group consisting of ammine; amine; peptide; halogen (chloride, fluoride or iodide); nucleobase; imidazole; H₂O; hydrogen; saturated or unsaturated alkyl, alcohol, or carboxylate of 1 to 10 carbon atoms; aromatic; or heterocycle of up to four rings.

Other preferred compounds include cobalt(II) phthalocyanine β -form; (S,S)-(+)-N-N'-bis(3,5-di-tert-butylsalicylidene)-1,2-cyclohexane diamino manganese (III)

chloride; iron(II) phthalocyanine bis(pyridine) complex;
iron(III) phthalocyanine chloride; manganese(II)
phthalocyanine; 5,10,15,20-tetraphenyl-21H, 23H-porphine
manganese (III) chloride; chloro(pyridine) bis
5 (dimethylglyoximato) cobalt (III); N-N'-
bis(salicylidene)dianilino-cobalt(II); cis-bis(2-2'-
bipyridine)dichloro-ruthenium (II) hydrate; and
cobalt(acacen)(NH₃)₂Cl, in which (H₂acacen represents
bis(acetylacetone)-ethylenediamine).

10 Compounds which are designed or selected according to
the methods of the invention may be tested for inhibitory
activity by any suitable assay procedure. Assays to
determine the binding of metal complexes to A β may be
performed by NMR or UV-Visible spectroscopy, or by ESR in
15 the case of paramagnetic metals. Assays are available for
measuring Cu/Fe reduction, hydrogen peroxide, hydroxyl
radical generation, and carbonyl group, all of which assess
the redox capacity of A β in the presence of Cu and Fe. *Ex*
vivo assays using *post mortem* brain tissue may also be
20 performed. These include measuring the amount of A β which
is solubilised and extracted in the presence of the
compound, and determining the quantity of peroxide formed
in *post mortem* brain tissue, as compared with control
tissue which is solubilised and extracted in the absence of
25 the compound. Suitable methods are described for example in
PCT/US99/05291 (WO99/45907).

The invention will now be described in detail by way
of reference only to the following non-limiting examples
and to the drawings.

30

Materials and Methods

NMR spectroscopy

Samples of A β 1-40 and A β 1-28 in aqueous solution were prepared by dissolving 0.5mgs and 1.0 mgs respectively in
5 0.55 ml of 10 % $^2\text{H}_2\text{O}/90\%\text{H}_2\text{O}$, 100 mM KCl and 50 mM phosphate buffer. pH values were measured at room temperature, and were not corrected for isotope or solvent effects. The ^1H chemical shifts were referenced to 2,2-dimethyl-2-silapentane-5-sulphonate at 0 ppm, via the
10 chemical shift of the H_2O resonance (Wishart et al., 1995a) or an impurity at 0.15 ppm.

Spectra were recorded on a Bruker DRX-600 spectrometer. All spectra in aqueous solution were recorded at 271 K and pH 6.8, with probe temperatures calibrated
15 according to the method of van Geet (1970). All 2D spectra were recorded in phase-sensitive mode using time-proportional phase incrementation (Marion & Wuthrich, 1983). Water suppression was achieved using pulsed field gradients with the WATERGATE scheme and a 3-9-19 selective
20 pulse (Sklenar et al., 1993).

2D homonuclear NOESY spectra (Anil-Kumar et al., 1980; Macura et al., 1981) were recorded with mixing times of 50 and 250 ms. TOCSY spectra (Braunschweiler & Ernst, 1983) were recorded using the DIPSI-2 spin-lock sequence
25 (Rucker & Shaka, 1989) with spin-lock times of 70-80 ms. DQF-COSY (Rance et al., 1983) spectra were also recorded. Typically, spectra were acquired with 400-600 t_1 increments, 32-128 scans per increment, and 4096 data points. The ^1H sweep width was 7575.6 Hz at 600 MHz.
30 Spectra were processed using UXNMR-941001.4 (Bruker) and analyzed using XEASY 1.3.7 (Bartels et al., 1995). Sine-squared window functions, phase shifted by 60° - 90° , were applied in both dimensions prior to Fourier transformation.

35 The $^3\text{J}_{\text{NHCOH}}$ coupling constants were measured from a DQF-COSY spectrum or by using the INFIT module of XEASY to analyze NOESY spectra.

For $^3J_{\text{NHCOH}}$ coupling constants measured from a DQF-COSY the appropriate rows were extracted from the spectrum, inverse Fourier transformed, zero filled to 32 K data points, and multiplied by a Gaussian window function prior to Fourier transformation. The dispersive peak shapes were simulated to take account of the effect of broad linewidths on small coupling constants, using an in-house program, COUPLING.

10 ***Metal Binding Studies***

Metal binding studies were performed by titrating concentrated metal solutions (30 mM CuCl_2 , ZnCl_2 in water) into the peptide solutions described above. The displacement of bound Cu^{2+} from A β 1-28 by the Co(III) Schiff-base was performed by adding two equivalents of Cu^{2+} to A β 1-28, followed by one equivalent of Co(III) Schiff-base.

Brain Tissue Assays

20 ***Tissue selection***

Post-mortem tissues, stored at -80°C , were obtained from the NH&MRC-supported Brain Bank at the University of Melbourne, together with accompanying histopathological and clinical data. AD was assessed according to CERAD criteria (Mirra et al, 1991). In order to examine the chemical architecture of the A β deposition observed in non-AD aged brain, A β immunohistochemistry was used to select age-matched control (AC) cases which did not fulfil the CERAD criteria, and in which amyloid deposition, if present, was detectable only in the form of diffuse plaques, but not neuritic plaques.

Preparation of compounds

Compounds were dissolved in DMSO and diluted in a PBS mixture. Insoluble material was removed.

Sample preparation:

The cortical meninges were removed and gray matter (0.5 g) was homogenised using a DIAX 900 homogeniser (Heidolph & Co, Kelheim, Germany) for 3 x 30s periods at full speed, with a 30s rest between strokes, in 3 ml of ice-cold phosphate-buffered saline (PBS), pH 7.4, containing a mixture of protease inhibitors (BioRad, Hercules, CA), but without ethylene diamine tetraacetic acid (EDTA), or in the presence of test compounds or metal ions prepared in PBS. The homogenate samples were incubated for 24 h at room temperature. To obtain the PBS-extractable fraction, the homogenate was centrifuged at 100,000 x g for 30 min, the supernatant removed, and divided into 1 ml aliquots. Protein in a 1 ml supernatant sample was precipitated using 1:5 ice-cold 10% trichloroacetic acid (TCA), and pelleted by centrifugation at 10,000 x g for 20 mins. The pellet was prepared for PAGE by boiling for 10 min in Tris-tricine SDS-sample buffer containing 8% SDS, 10% mercaptoethanol and 8M urea. Total A β in the cortical samples was obtained by homogenizing in 1 ml PBS and boiling in sample buffer as described above.

Polyacrylamide gel electrophoresis (PAGE) and Western blotting

Tris-tricine PAGE was performed by loading samples on to 10-20% gradient gels (Novex, San Diego, CA), followed by transfer on to 0.2 mm nitrocellulose membrane (BioRad, Hercules, CA). The A β was detected using the following monoclonal antibodies: W02, which detects A β 40 and A β 42 at an epitope between residues 5 and 8; G210, which is specific for A β species which terminate at carboxyl residue 40; or G211, which is specific for A β species which terminate at carboxyl residue 42 (Ida et al, 1996), in conjunction with horseradish peroxidase(HRP)-conjugated rabbit anti-mouse IgG (Dako, Denmark), and visualised using chemiluminescence (ECL, Amersham Life Science, UK). Each gel included two or more lanes containing known quantities of synthetic A β (Keck Laboratory, Yale University New

Haven, CT) as reference standards. Blot films were scanned using a Relisys scanner with transparency adapter (Tech Information Systems, Taiwan) and densitometry performed using Image 1.6 software (NIH, Bethesda, MA). All samples
5 were analysed at least twice, and gel loadings and dilutions adjusted to fit within the quantifiable region of the standard curve.

The efficiency of the TCA precipitation procedure was validated by testing samples of whole human serum diluted
10 1:10 to which had been added 2mg of synthetic A β 1-40 or A β 1-42. A β recovery was assessed by extracting the precipitate into SDS sample buffer and performing Western blot analysis, using synthetic A β standards as described above. Protein in the TCA pellet was estimated by
15 resuspending the pellet in water and assaying the protein recovery using a BCA assay (Pierce, Rockford, IL). This indicated that the efficiency of protein and A β precipitation was approximately 90%. The efficiency of the 8M urea solubilization was found to be equivalent to that
20 of formic acid in a parallel, blinded assay conducted independently. All chemicals were obtained from Sigma (St. Louis, MO) unless otherwise indicated.

25 **Example 1 Characterisation of A β peptides in aqueous solution**

In aqueous solution there is little chemical shift difference between the amide and C $^{\alpha}$ H protons of A β 1-28 compared with A β 1-40, suggesting that both peptides are in a similar conformation. Comparisons of A1-28 and A β 1-40
30 chemical shifts with random coil chemical shifts and the lack of NOE connectivities in the NOESY spectra indicate that both peptides are mostly in conformational exchange. However, there are some medium range NOE connectivities (1 < /ii-j/ < 5) observed in the region of residues 16-21 of
35 the peptide (KLVFFA), suggesting that this region of the peptide has a structured turn. This is illustrated in Figure 1.

This region of the peptide has previously been shown to be very important in defining the aggregation properties of A β (Hilbich et al. 1992), with the substitution of hydrophilic residues into this region resulting in altered aggregation properties, including reduced β -sheet content. In addition, several groups have described short peptides or slight variants thereof corresponding to this region which have the ability to bind to A β and to inhibit the formation of amyloid fibrils (Findeis et al. 1999; Tjernberg et al. 1999). This evidence implies that this "structured" section of A β is important in the formation of amyloid fibrils.

Example 2

Metal Binding Studies

To determine the metal-binding site of A β 1-40, Zn²⁺ was titrated into a solution of A β 1-40 in SDS-micelles at pH 6.5. Peaks due to the C2H protons of the imidazole rings of His6, His13 and His14 broadened out such that they were no longer visible when a small amount of Zn solution (~ 25% of one mol. equivalents) was added. The addition of extra Zn (up to two mol. equivalents) did not change the spectrum, but when the pH of the solution was raised to 7.4 three broad overlapping peaks due to the C2H protons of the imidazole rings of His6, His13 and His14 became visible. These peaks did not sharpen significantly even upon the addition of a large excess of Zn (> 150 mol. equivalents). There appear to be no significant difference in the rest of the spectrum between the Zn-bound and free forms of A β 1-40, suggesting that there are no significant conformational changes upon metal binding. These results indicate that all three histidine residues of A β 1-40 are involved in Zn binding. Figure 2 shows a model of Zn bound to the three histidine residues of A β 1-40.

To determine the metal-binding site of A β 1-40 and A β 1-28 in aqueous solution, Zn²⁺ and Cu²⁺ were titrated into solutions of A β 1-40 and A β 1-28 at pH 6.9. All reactions were accompanied by significant precipitation. The NMR

spectrum of the peptide-metal complex which remained in solution showed that peaks due to the C2H and C4H protons of His6, His13 and His14 broadened out such that they were no longer visible, indicating that these residues were involved in metal binding. This is illustrated in Figures 3 and 4. The addition of more metal ion resulted in more precipitation, so that saturated binding was not possible.

When the Co(III) Schiff-base complex was added to a solution containing Cu²⁺-bound A β 1-28, a broad peak appeared in the 1H spectrum at 6.55 ppm, as shown in Figure 5. The chemical shift of this peak is consistent with the chemical shift of a C4H proton of a histidine imidazole bound to a Co(III)Schiff-base complex, as shown in Figure 6. This indicates that the Co(III)Schiff-base complex can compete with Cu²⁺ for the histidine residues of A β .

Example 3 Brain tissue assays

Brain tissue assays were conducted in order to test the ability of the following compounds to reduce β -amyloid peptide aggregation:

- | | |
|--------|--|
| KJB001 | Co(II) phthalocyanine β -form |
| KJB002 | (S,S)-(+) -N-N'-bis(3,5-di-tert-butylsalicylidene)-1-2-cyclohexane diamino |
| 25 | manganese (III) chloride |
| KJB003 | Haemin |
| KJB004 | Iron(II) phthalocyanine bis(pyridine) complex |
| KJB005 | Iron(III) phthalocyanine chloride |
| KJB006 | Manganese(II) phthalocyanine |
| 30 | KJB007 5,10,15,20-tetraphenyl-21H,23H-porphine |
| | manganese(III) chloride |
| KJB008 | Chloro(pyridine) bis(dimethylglyoximato)cobalt (III) |
| KJB009 | N-N'-bis(salicylidene)dianilino-cobalt(II) |
| 35 | KJB010 cis-bis(2-2'-bipyridine)dichloro-ruthenium (II) hydrate. |

BR16805 [Co(acacen)(NH₃)₂]Cl (H₂acacen represents
bis(acetylacetone)ethylenediamine)

Table 1 shows the approximate concentration of each
5 compound used in the extraction.

Table 1

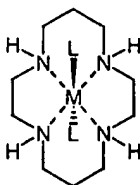
Compound	Concentration (μ mol/litre)
KJB001	0.0105
KJB002	4.41
KJB003	60.87
KJB004	3.816
KJB005	3.08
KJB006	111.5
KJB007	0.64
KJB008	1289.9
KJB010	300.9
BR16805	1199.9
bathocuproine	2000

10 The results presented in Figure 7 clearly demonstrate
that some of the metal complexes have the ability to
solubilise A β deposited in the brains of Alzheimer's
disease patients. In particular, compound KJB001 showed
particularly strong solubilising activity, and compounds
15 KJB002, KJB005, KJB006, KJB007 and KJB010 also showed good
activity.

Example 4 Compounds designed to bind one or more of
the three histidine residues in A β

20 BRI7080, BRI7103 and BRI7104 are metal complexes of
aza-macrocycles of the following structures.

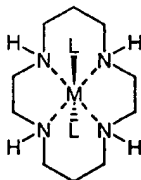
(a) BRI7080



BRI7080 : $M = Ni^{2+}$ $L = H_2O$

Ni(NO₃)₂·6H₂O (1.67 g, 5.75 mmol) was added in portions to a solution of 1,4,8,11-tetraazacyclotetradecane (1.0 g, 5.00 mmol) in ethanol (50 ml) at 50 °C. The mixture was heated at 50 °C for 2 h during which time the green coloured nickel salt dissolved to give a red solution. Upon cooling of the crude reaction mixture to room temperature, a purple solid had precipitated which was collected by filtration and washed with several portions of 2-propanol. The product thus obtained was dried overnight in a vacuum oven. Spectral and analytical data acquired for this compound were in agreement with literature data.

(b) BRI7103



BRI7103 : $M = Mn^{3+}$ $L = Cl$

BRI7103 was prepared following the method of P-K. Chan and C-K. Poon, *J. C. S. Dalton Trans.*, 1976, 858-862.

(c) BRI7104

This compound has similar structures to BRI7080 and BRI7103, but in this case $M = Co^{3+}$ and $L = Cl$.

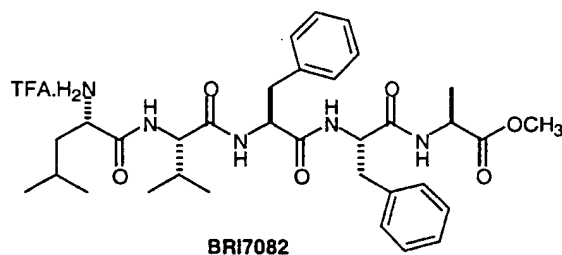
BRI7104 was prepared following the method of P-K. Chan and C-K. Poon, *J. C. S. Dalton Trans.*, 1976, 858-862.

Example 5 **Histidine binding compounds conjugated with an A β -directing group.**

For the second class of molecules we chose derivatives of the pentapeptide leucine-valine-phenylalanine-phenylalanine-alanine (LVFFA), designated herein as BRI7082 and BRI7077, to direct histidine binding compounds to A β .

BRI7106 is an example of a nicotine derivative conjugated to an A β -directing group. In this example the histidine-binding group is a small organic molecule rather than a metal complex, in contrast to most of the compounds described above. BRI7158 and BRI7159 are examples of metal complexes conjugated to an A β -directing group.

(a) BRI7082



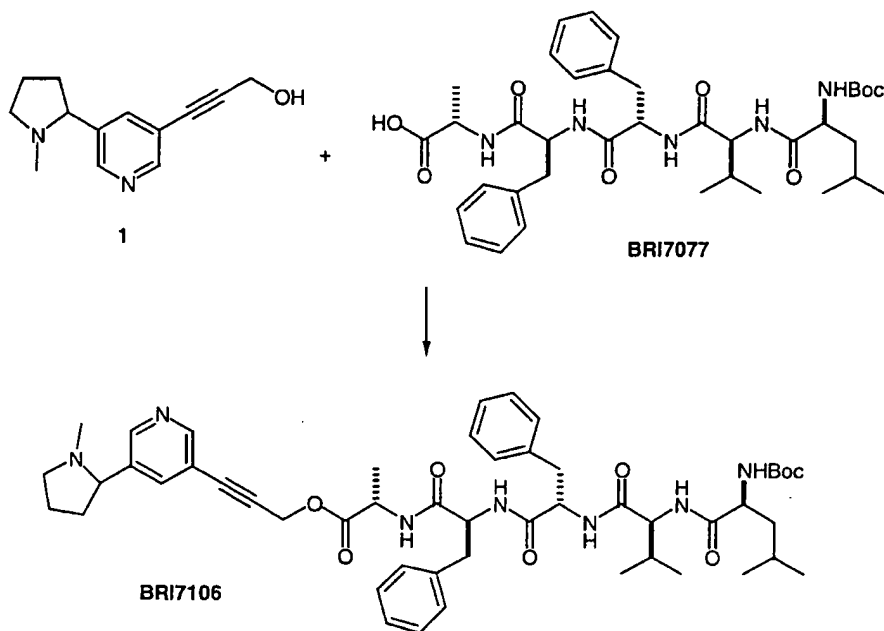
BRI7082 was prepared via standard HBTU-mediated peptide coupling methodology; see M. Bodanszky and A. Bodanszky, *"The Practice of Peptide Synthesis"*, 2nd Edition, Springer-Verlag, 1994.

Mass Spectrum (APCI) m/z 610 [(MH-C₂H₂O₂F₃)⁺, 100%].

25

(b) BRI7106

This compound was synthesised according to Reaction Scheme 1:



Scheme 1

Compound 1 was prepared following the methods of P. Jacob III, *J. Org. Chem.*, 1982, 47, 4165-4167 and N. D. P. Cosford et al., *J. Org. Chem.*, 1998, 63, 1109-1118.

5 Mass Spectrum (APCI) m/z 217 [(M+H)⁺, 100%].

BRI7077 was prepared via standard HBTU-mediated peptide coupling methodology; see M. Bodanszky and A. Bodanszky, *"The Practice of Peptide Synthesis"*, 2nd Edition, Springer-Verlag, 1994.

10 Mass Spectrum (APCI) m/z 696 [(M+H)⁺, 20%], 694 [(M-H), 95%].

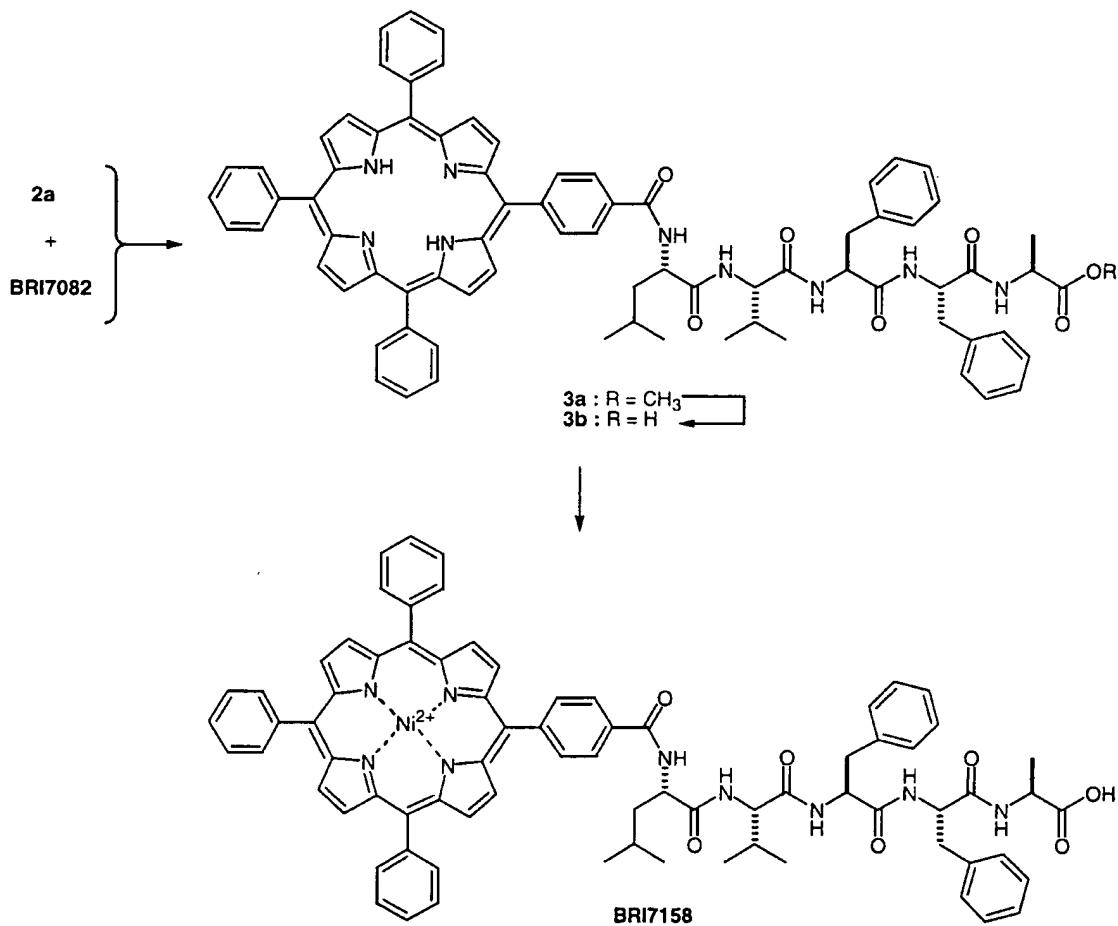
BRI7106 was prepared via standard DCC-mediated esterification methodology; see M. Bodanszky and A. Bodanszky, *"The Practice of Peptide Synthesis"*, 2nd

15 Edition, Springer-Verlag, 1994.

Mass Spectrum (APCI) m/z 894 [(M+H)⁺, 10%], 892 [(M-H), 25%].

(c) Synthesis of BRI7158

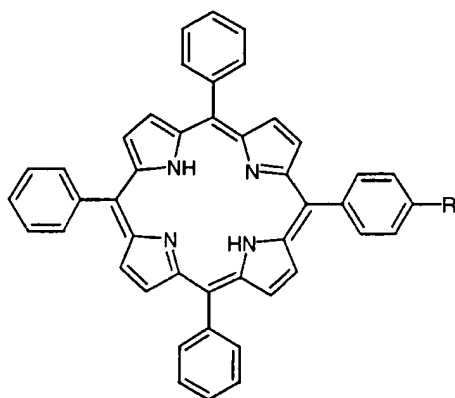
This compound was synthesised according to Reaction Scheme 2:



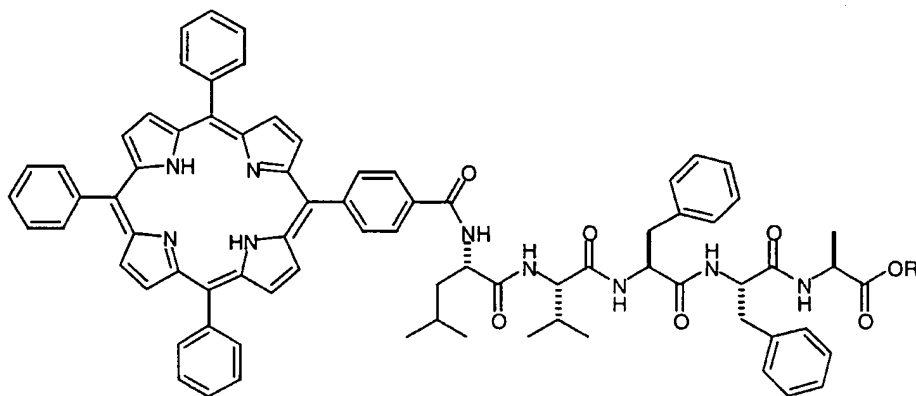
Scheme 2

5

Porphyrin 2a was prepared following the methods of K. Nakanishi et al., *Heterocycles*, 1996, 42, 723-736 and D. A. James et al., *Bioorg. Med. Chem. Lett.*, 1999, 9, 2379-2384.

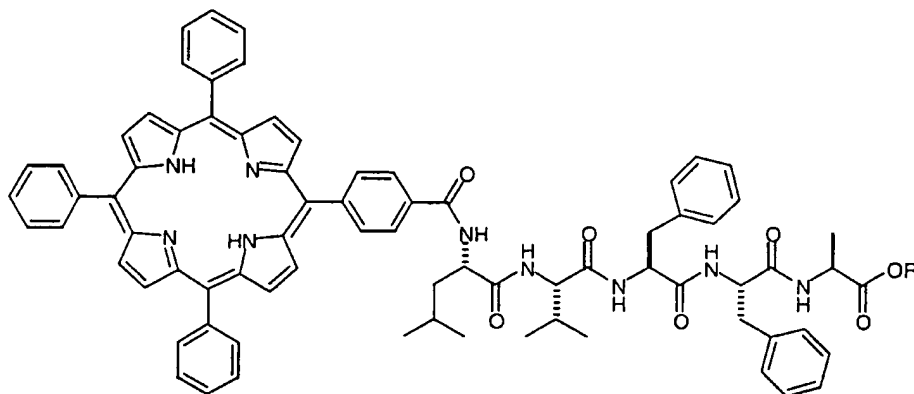
2a : R = CO₂H

N,N-Diisopropylethylamine (0.10 ml, 0.57 mmol) was added to a magnetically stirred mixture of porphyrin 2a (118 mg, 0.18 mmol), HBTU (68 mg, 0.18 mmol), and BRI7082 (130 mg, 0.18 mmol) in dry DMF (4 ml) under an atmosphere of nitrogen. Stirring was continued at room temperature for 16 h, after which time, the crude reaction mixture was partitioned between ethyl acetate and brine. The separated organic layer was successively washed with 10% aqueous citric acid solution, saturated aqueous NaHCO₃ solution and brine before being dried (MgSO₄), filtered and concentrated under reduced pressure to afford compound 3a as a deep purple, crystalline solid.

3a : R = CH₃

Mass Spectrum (APCI) m/z 1250 $[(M+H)^+]$, 45%].

A solution of compound 3a (23.3 mg, 0.019 mmol, THF (0.5 ml), methanol (50 drops) and 2M aqueous NaOH solution (0.1 ml) was stirred at room temperature for 18 h. The crude reaction mixture was partitioned between ethyl acetate and water and the phases separated. The aqueous layer was acidified to pH 4 upon the addition of 10% aqueous citric acid solution and then extracted three times with ethyl acetate. The combined organic layers were dried ($MgSO_4$), filtered and concentrated under reduced pressure to afford compound 3b as a purple solid.



3b : R = H

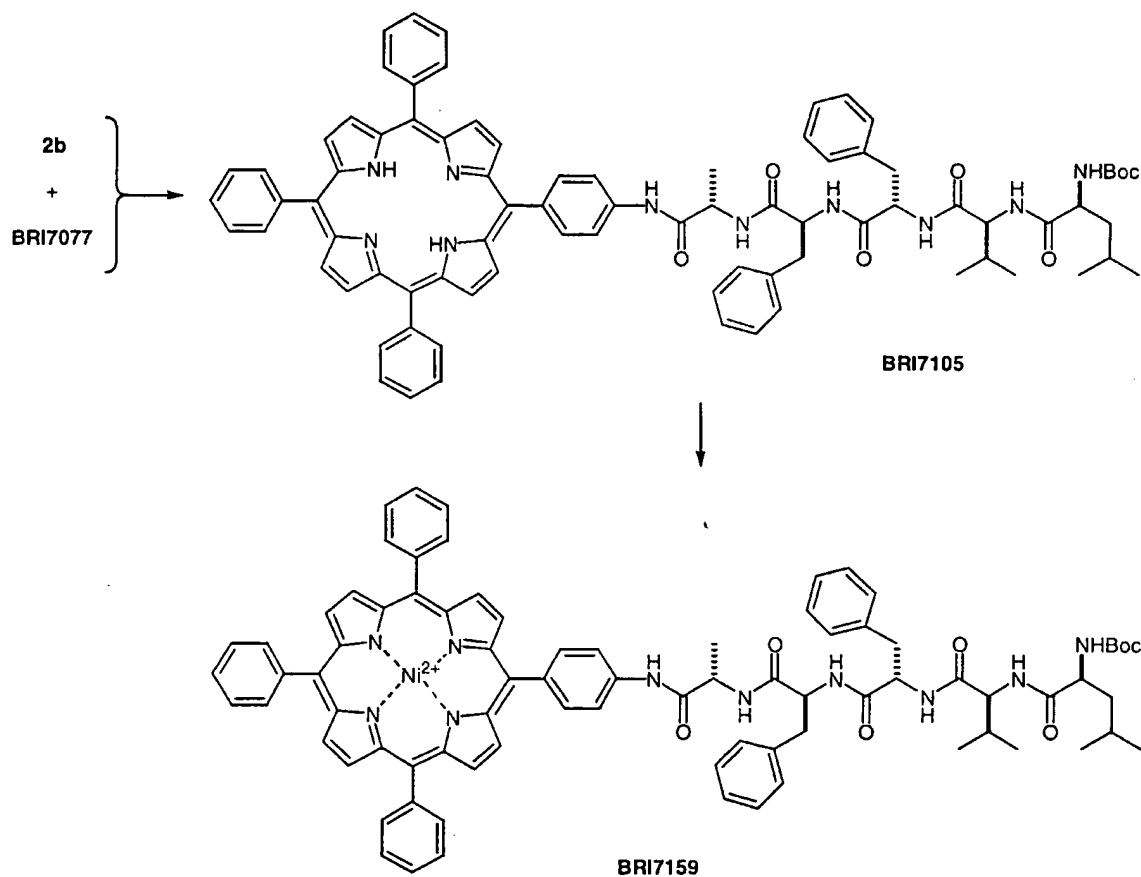
Mass Spectrum (APCI) m/z 1236 $[(M+H)^+]$, 30%].

A magnetically stirred mixture of compound 3b (17 mg, 0.014 mmol), $Ni(OAc)_2 \cdot 4H_2O$ (34.2 mg, 0.14 mmol), glacial acetic acid (0.73 ml) and dichloromethane (1.45 ml) was heated at reflux under an atmosphere of nitrogen for 18 h. The now crimson red reaction mixture was cooled to room temperature and partitioned between ethyl acetate and 10% aqueous $NaHCO_3$ solution. The separated aqueous phase was extracted three times with ethyl acetate before being dried (Na_2SO_4), filtered and concentrated under reduced pressure to give a red solid. Subjection of this material to flash chromatography (silica, 10% methanol/dichloromethane

elution) afforded, after concentration of the appropriate fractions (R_f 0.30), BRI7158 as a red, crystalline solid. Mass Spectrum (ES) m/z 1291 [(M-H), <10%].

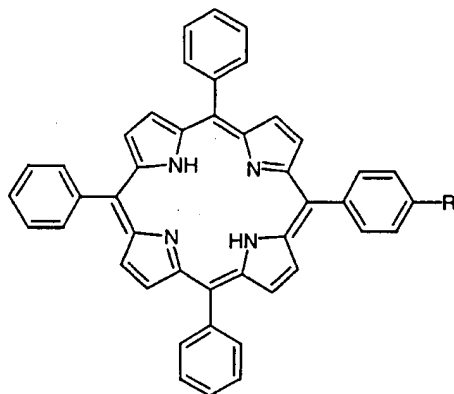
5 (d) Synthesis of BRI7105 and BRI7159

These compounds were synthesised according to Reaction Scheme 3:



Scheme 3

Porphyrin 2b was prepared following the method of W. J. Kruper, Jr. et al., *J. Org. Chem.*, 1989, 54, 2753-2756.



2b : R = NH₂

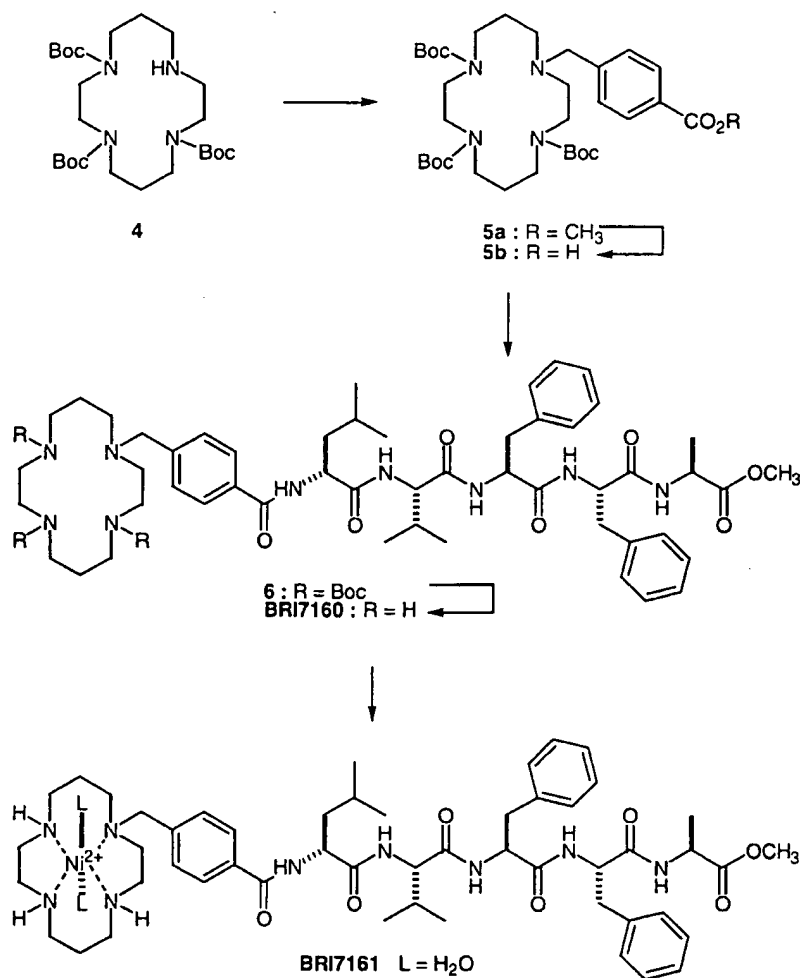
5 *N,N*-Diisopropylethylamine (0.21 ml, 1.18 mmol) was added at room temperature to a magnetically stirred mixture of porphyrin 2b (151 mg, 0.24 mmol), HBTU (216 mg, 0.58 mmol), and BRI7077 (200 mg, 0.29 mmol) in dry DMF (2 ml) under an atmosphere of nitrogen. Stirring was continued at 40°C for 40 h, after which time the crude reaction mixture
10 was partitioned between ethyl acetate and brine. The separated organic layer was successively washed with 10% aqueous citric acid solution, 5% aqueous NaHCO₃ solution and brine before being dried (MgSO₄), filtered and concentrated under reduced pressure to afford a deep purple
15 glass. Subjection of this material to flash chromatography (silica, 1% to 5% methanol/dichloromethane elution) gave, after concentration of the relevant fractions (*R_f* 0.35), BRI7105 as a purple, crystalline solid.
Mass Spectrum (APCI) *m/z* 1307 [(*M*+*H*)⁺, <5%].

20 A magnetically stirred mixture of BRI7105 (47 mg, 0.04 mmol), Ni(OAc)₂·4H₂O (96 mg, 0.39 mmol), glacial acetic acid (2 ml) and dichloromethane (4 ml) was heated at reflux under an atmosphere of nitrogen for 18 h. The now crimson red reaction mixture was cooled to room temperature
25 and partitioned between ethyl acetate and 10% aqueous

NaHCO₃ solution. The separated aqueous phase was extracted three times with ethyl acetate before being dried (Na₂SO₄), filtered and concentrated under reduced pressure to give a red solid. Subjection of this material to flash chromatography (silica, 10% methanol/dichloromethane elution) afforded, after concentration of the appropriate fractions (*R_f* 0.30), BRI7159 as a red, crystalline solid. Mass Spectrum (APCI) *m/z* 1385 [(*M*+Na)⁺, <5%].

10 (e) Synthesis of BRI7160 and BRI7161

The compounds were prepared according to Reaction Scheme 4:



Scheme 4

Compound 4 was prepared following the method of R. Guillard et al., *Bull. Soc. Chim. Fr.*, 1996, 133, 65-73.

A magnetically stirred suspension of compound 4 (430 mg, 0.86 mmol), (4-bromomethyl) methyl benzoate (217 mg, 0.95 mmol), KHCO_3 (172 mg) and K_2CO_3 (174 mg) in dry acetonitrile (17 ml) was heated at reflux for 18 h under an atmosphere of nitrogen. The reaction mixture was cooled to room temperature then concentrated under reduced pressure to yield a straw coloured oil which was purified by flash chromatography (silica, 5% methanol/dichloromethane elution). Concentration of the appropriate fractions (R_f 0.38) afforded adduct 5a as a clear, colourless oil. Mass Spectrum (APCI) m/z 649 [$(M+H)^+$, 100%].

A mixture of compound 5a (120 mg, 0.19 mmol), 1M aqueous LiOH (5 ml) and THF (5 ml) was magnetically stirred at room temperature for 16 h. The crude reaction mixture was partitioned between ether and water and the separated aqueous layer was cooled to 0°C prior to acidification to pH 4 with 10% aqueous citric acid solution. Extraction of the aqueous layer with ethyl acetate, followed by drying of the organic fraction with MgSO_4 afforded, after concentration under reduced pressure, compound 5b as a viscous, colourless oil.

Mass Spectrum (APCI) m/z 633 [$(M-H)$, 100%].

N,N-Diisopropylethylamine (0.17 ml, 0.98 mmol) was added to a magnetically stirred mixture of compound 5b (167 mg, 0.26 mmol), HBTU (132 mg, 0.35 mmol), and BRI7082 (209 mg, 0.29 mmol) in dry DMF (2 ml) under an atmosphere of nitrogen. Stirring was continued at room temperature for 16 h, after which time, the crude reaction mixture was partitioned between ethyl acetate and brine. The separated organic layer was successively washed with 10% aqueous citric acid solution, saturated aqueous NaHCO_3 solution and brine before being dried (MgSO_4), filtered and concentrated under reduced pressure to afford compound 6 as a viscous oil.

Mass Spectrum (APCI) m/z 1226 $[(M+H)^+]$, 65%].

A solution of compound 6 (258 mg, 0.21 mmol) in dichloromethane (4 ml) was treated with trifluoroacetic acid (0.1 ml) under an atmosphere of nitrogen and stirring
5 was continued at room temperature for 20 h. The crude reaction mixture was concentrated under reduced pressure to afford a light tan oil which was purified by flash chromatography (silica, 6% methanol/dichloromethane elution). Concentration of the relevant fractions (R_f
10 0.30) afforded BRI7160 as a pale yellow foam.

Mass Spectrum (APCI) m/z 633 $[(M-H)]$, 100%].

A solution of BRI7160 (41 mg, 0.044 mmol) and $Ni(OAc)_2 \cdot 4H_2O$ (103 mg, 0.41 mmol) in dry methanol (2 ml) was heated at reflux for 19 h under an atmosphere of
15 nitrogen. The cooled reaction mixture was concentrated under reduced pressure to afford a solid residue. This material was treated with chloroform (4 ml) and the resulting suspension was filtered. Concentration of the filtrate afforded BRI7161 as a foam-like residue.

20 Mass Spectrum (APCI) m/z 1020 $[(M+H)^+]$, 30%].

Example 5 Testing of compound KJB030 for ability to inhibit amyloid β -peptide neurotoxicity in vitro

Compound KJB030 was dissolved in chelex 100-treated
25 double distilled H_2O (CH100-d H_2O) at a concentration of 200 μM . A β 1-42 was dissolved in either 200 μM KJB030 solution or in CH100-d H_2O . Both solutions contained 200 μM A β 1-42. The A β 1-42 solutions were incubated at ambient temperature (~20-24°C) for 3 days. The solutions were then brought to a
30 physiological state by adding 10x phosphate buffered saline (PBS), pH 7.4, thus bringing samples to 1x PBS. Samples were then incubated at 37°C for 24 hr to induce aggregation of A β 1-42. Samples were subsequently dialysed in microdialysis cups (3500 M_r cut-off) placed in 5 L PBS for
35 24 hr, in order to remove some of the excess KJB030 without loss of A β 1-42.

Dialyzed samples were added to six day old cerebral cortical neuron cultures growing in Neurobasal medium with B27 supplements but without anti-oxidants (Gibco). Samples were diluted into the culture medium to a final

5 concentration of 20 μ M A β 1-42. The concentration of the KJB030 was unknown, due to dialysis of the sample; however, the maximum concentration would be 20 μ M if no dialysis had occurred. Six wells of each of the following were established: 1; Untreated control cells, 2; A β 1-42 (20 μ M),

10 3; KJB030 (dialysed), 4; KJB030+A β 1-42 (20 μ M, dialysed), 5; vehicle only (PBS). Cultures were treated for 5 days (37°C/5%CO₂) and culture medium was removed for analysis of cell death.

Determination of cell death was performed using the

15 lactate dehydrogenase (LDH) assay kit (Boehringer Mannheim). This is a measure of the release of LDH, a constitutively expressed, highly abundant enzyme, from dead (lysed) cells. Culture medium was mixed with the reagents from the LDH kit as per the manufacturer's instructions and

20 the level of LDH measured colourimetrically on a spectrophotometric 96 well plate reader at 490 nm. Using the equation supplied with the LDH assay kit, the percentage of LDH release and hence cell death compared to untreated control cultures was determined. The results are

25 shown in Table 1.

Table 1

Results of LDH assay for cell death

Treatment:	% cell death
Untreated control neurons:	0
Vehicle alone (PBS)	-1.18 \pm 1.4
KJB030 alone	8.1 \pm 2.2
A β 1-42 (20 μ M) alone	22 \pm 3.5
A β 1-42 (20 μ M) + KJB030	3.9 \pm 2.5*

*Significantly different from A β 1-42 alone (p<0.01)

Example 6 **Demonstration of Reaction of Compound**
KJB001 with A β 1-28

1.8 mgs of A β 1-28 was dissolved in 5 mls DMSO was
5 mixed with 0.33 mgs of compound KJB001 (~1 equivalent)
dissolved in 0.3 mls DMSO. The mixture was allowed to
stand for 24 hrs before being freeze-dried. The resulting
blue powder was dissolved in 550 μ l of an aqueous solution
containing 100 mM NaCl, 50 mM phosphate buffer, pH 6.9. A
10 solution of of 1.8 mgs A β 1-28 dissolved in 550 μ l of the
same solution was used as a control.

Figure 9(A) shows the 600 MHz ^1H NMR spectrum of the
aqueous solution of A β 1-28 at 271 K. The resonances due to
the C2H protons of His6, 13, 14 are marked with *. Figure
15 9(B) shows the ^1H 600 MHz NMR spectrum of the solution of
A β 1-28 plus KJB001, demonstrating that the peaks due to the
histidine C2H protons had shifted, thus indicating that
KJB001 had reacted with these residues.

20 **DISCUSSION**

When copper and iron bind to A β , reactive oxygen
species such as peroxide and superoxide are produced. When
copper and zinc bind to A β , both induce aggregation, and
copper binding is inhibited by zinc, suggesting that these
25 ions bind to similar binding sites, or share a single
binding site. Zinc, and presumably copper, bind to the
histidine residues of A β . Thus a molecule which prevents
the binding of zinc and copper to these histidine residues
has the potential to inhibit A β aggregation and to prevent
30 metal-induced neurotoxicity.

Compounds of the kind described herein have the
potential to bind to histidine residues and therefore to
prevent zinc and copper binding, and so may have
therapeutic value. A model of a cobalt-corrin ring complex
35 bound to A β 1-40 is shown in Figure 8.

It will be apparent to the person skilled in the art that while the invention has been described in some detail for the purposes of clarity and understanding, various modifications and alterations to the embodiments and
5 methods described herein may be made without departing from the scope of the inventive concept disclosed in this specification.

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10

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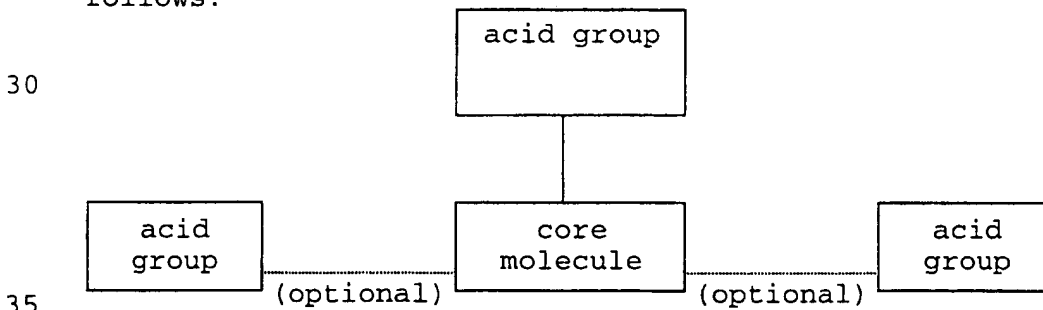
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CLAIMS

1. A compound which interacts with the β -amyloid peptide in such a way that the N-terminal loop of the peptide
5 (amino acid residues 1-15) is blocked or destabilised, thereby inhibiting the binding of one or more metal ions to at least one histidine residue within the N-terminal loop.
2. A compound according to claim 1 which inhibits binding of Cu^{2+} , Zn^{2+} and Fe^{3+} ions, but not Mg^{2+} or Ca^{2+}
10 ions.
3. A compound according to claim 1 or claim 2 which has a conformation and polarity such that it binds to at least one histidine residue in the N-terminal loop, selected from the group consisting of His6, His13 and
15 His14.
4. A compound according to claim 3, which binds to at least two histidine residues in the N-terminal loop.
5. A compound according to claim 4, which binds to at least three histidine residues in the N-terminal loop.
- 20 6. A compound according to any one of claims 1 to 5, which also binds to at least one additional amino acid in the N-terminal loop, selected from the group consisting of Asp7, Tyr10, and Glu11.
7. A compound according to any one of claims 1 to 5, which has acidic groups which interact with one or more of
25 the His residues in the N-terminal loop.
8. A compound according to claim 7, represented as follows:



wherein the core molecule has a conformation and polarity such that the acid group(s) interact with one of more of His6, His13 and His14.

9. A compound according to claim 9, in which the acid group is selected from the group consisting of CO₂H, PO₃H₂, SO₃H, OSO₃H₂, and OPO₃H₂.
10. A compound according to claim 9, which is a molecule with one to three carboxylic acid groups, the length of the molecule being such that it can be received within the N-terminal loop, and such that at least one carboxyl group is in proximity to at least one of the histidine residues.
11. A compound according to any one of claims 1 to 10, which is an organic molecule, a peptide or a metal complex.
12. A compound according to claim 9, which is not a metal complex.
13. A compound according to claim 9, which has overall hydrophobic character.
14. A compound according to claim 10, which is able to penetrate the blood-brain barrier.
15. A compound according to any one of claims 1 to 14, which comprises, or is conjugated to, a targeting moiety.
16. A compound according to claim 15, in which the targeting moiety is selected from the group consisting of polypeptides, nucleic acids, carbohydrates, lipids, β -amyloid ligands, antibodies, and dyes.
17. A compound according to claim 15, in which the targeting moiety has a hydrophobic region which interacts with the tail of the β -amyloid peptide.
18. A compound according to claim 17, in which the targeting moiety comprises a fatty acid molecule.
19. A compound according to any one of claims 15 to 18, in which the targeting moiety targets the compound to the site defined by residues 15-21 of the β -amyloid peptide.
20. A compound according to claim 17, in which the targeting moiety is a peptide which comprises a sequence which corresponds to that of residues 15-21 of the β -amyloid peptide.

21. A compound according to any one of claims 15 to 20, in which the inhibitor-targeting moiety complex is able to penetrate the blood-brain barrier.

22. A method of selecting or designing a compound which
5 inhibits the binding of metal ions to the N-terminal loop of the β -amyloid peptide, which method comprises the steps of

(i) selecting or designing a compound which has a conformation and polarity such that it binds to at least
10 one, more preferably at least two and more preferably three amino acids in the N-terminal loop selected from the group consisting of His6, His 13 and His14; and

(ii) testing the compound for the ability to inhibit binding of metal ions to the N-terminal loop of the
15 β -amyloid peptide.

23. A method according to claim 22, in which the compound binds to at least two histidine residues in the N-terminal loop.

24. A method according to claim 23, in which the compound
20 binds to at least three histidine residues in the N-terminal loop.

25. A method according to any one of claims 22 to 24, in which the compound also binds to at least one additional amino acid in the N-terminal loop, selected from
25 the group consisting of Asp7, Tyr10, and Glu11. Preferably the compound inhibits binding of both Cu^{2+} and Zn^{2+} ions, but not Mg^{2+} or Ca^{2+} ions.

26. A method according to claim 26, in which the compound inhibits binding of Cu^{2+} , Zn^{2+} and Fe^{3+} ions, but not Mg^{2+} or
30 Ca^{2+} ions.

27. A method according to any one of claims 22 to 26, in which the compound has overall hydrophobic character.

28. A method according to claim 27, in which the compound is able to penetrate the blood-brain barrier.

29. A compound which inhibits the binding of metal ions
35 to the N-terminal loop of the β -amyloid peptide, wherein

the compound is obtained by a method according to any one of claims 22 to 28.

30. A composition comprising a compound according to any one of claims 1 to 21 or claim 29, together with a

5 pharmaceutically acceptable carrier.

31. A method of inhibiting the binding of one or more metal ions to the β -amyloid peptide, or of inhibiting the aggregation of β -amyloid peptide, which method comprises the step of exposing the peptide to a compound which blocks
10 or destabilises the N-terminal loop of the peptide, thereby inhibiting the binding of one or more metal ions to at least one histidine residue within the N-terminal loop.

32. A method according to claim 31, in which the compound has a conformation and polarity such that it binds to at
15 least one histidine residue in the N-terminal loop of the β -amyloid peptide, selected from the group consisting of His6, His13 and His14.

33. A method according to claim 32, in which the compound binds to at least two histidine residues in the
20 N-terminal loop.

34. A method according to claim 33, in which the compound binds to at least three histidine residues in the N-terminal loop.

35. A method according to any one of claims 31 to
25 34, in which the compound also binds to at least one additional amino acid in the N-terminal loop, selected from the group consisting of Asp7, Tyr10, and Glu11.

36. A method according to any one of claims 31 to 35, in which the compound inhibits binding of Cu^{2+} , Zn^{2+} and Fe^{3+}
30 ions, but not Mg^{2+} or Ca^{2+} ions.

37. A method according to any one of claims 31 to 36, in which the compound is a complex of Mn, Fe, Co, Ni, Cu, Zn, Ru, Pd, Ag, Cd, Pt, Au, Rh or Hg, with the proviso that the compound is not haemin or haematin.

38. A method according to any one of claims 31 to 37, in which the compound comprises, or is conjugated to, a
35 targeting moiety.

39. A method according to claim 38, in which the targeting moiety targets the compound to the site defined by residues 15-21 on the β -amyloid peptide.

40. A method according to any one of claims 31 to 39, in
5 which the inhibition of binding of one or more metal ions to the β -amyloid peptide occurs *in vivo*.

41. A method of prevention, treatment or alleviation of Alzheimer's disease, which method comprises the step of administering a compound according to any one of claims 1
10 to 21 or a pharmaceutical composition according to claim 30 to a subject in need of such treatment.

42. A method of prevention, treatment or alleviation of Alzheimer's disease, which method comprises inhibiting the binding of one or more metal ions to the β -amyloid peptide,
15 or inhibiting the aggregation of β -amyloid peptide, by the method of claim 40.

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Figure 1

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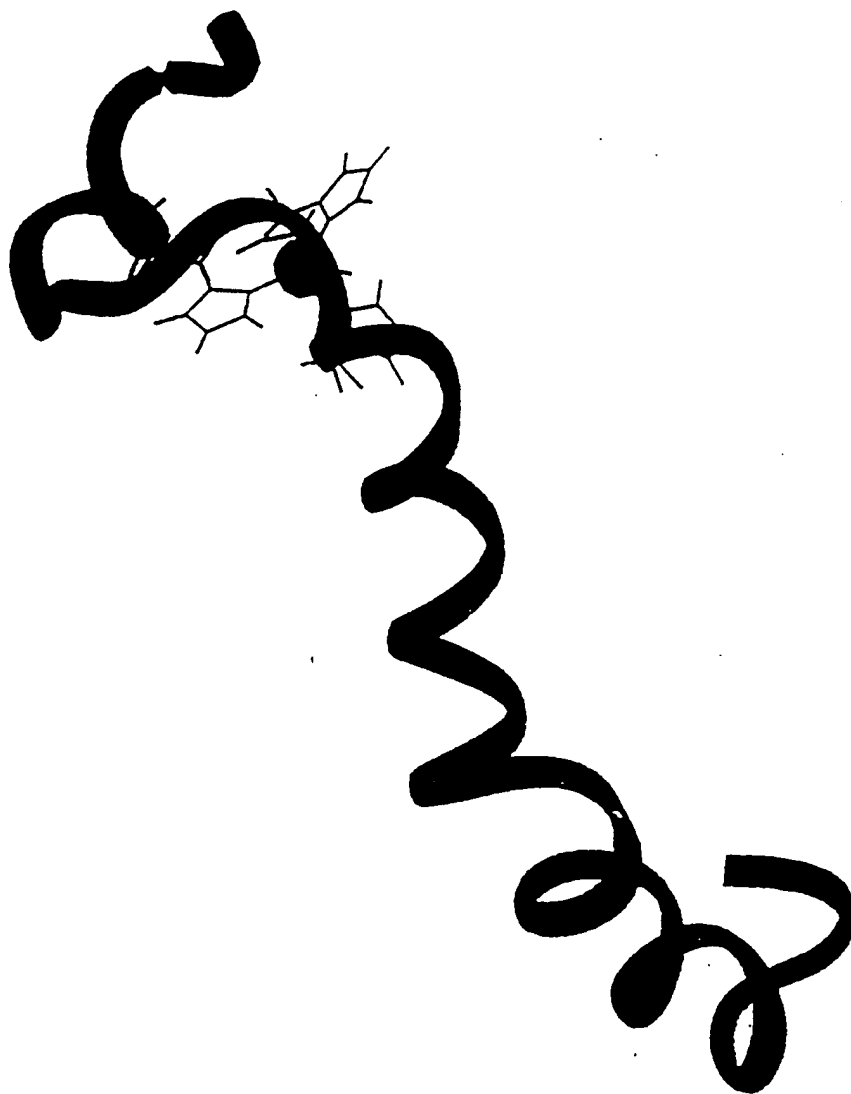


Figure 2

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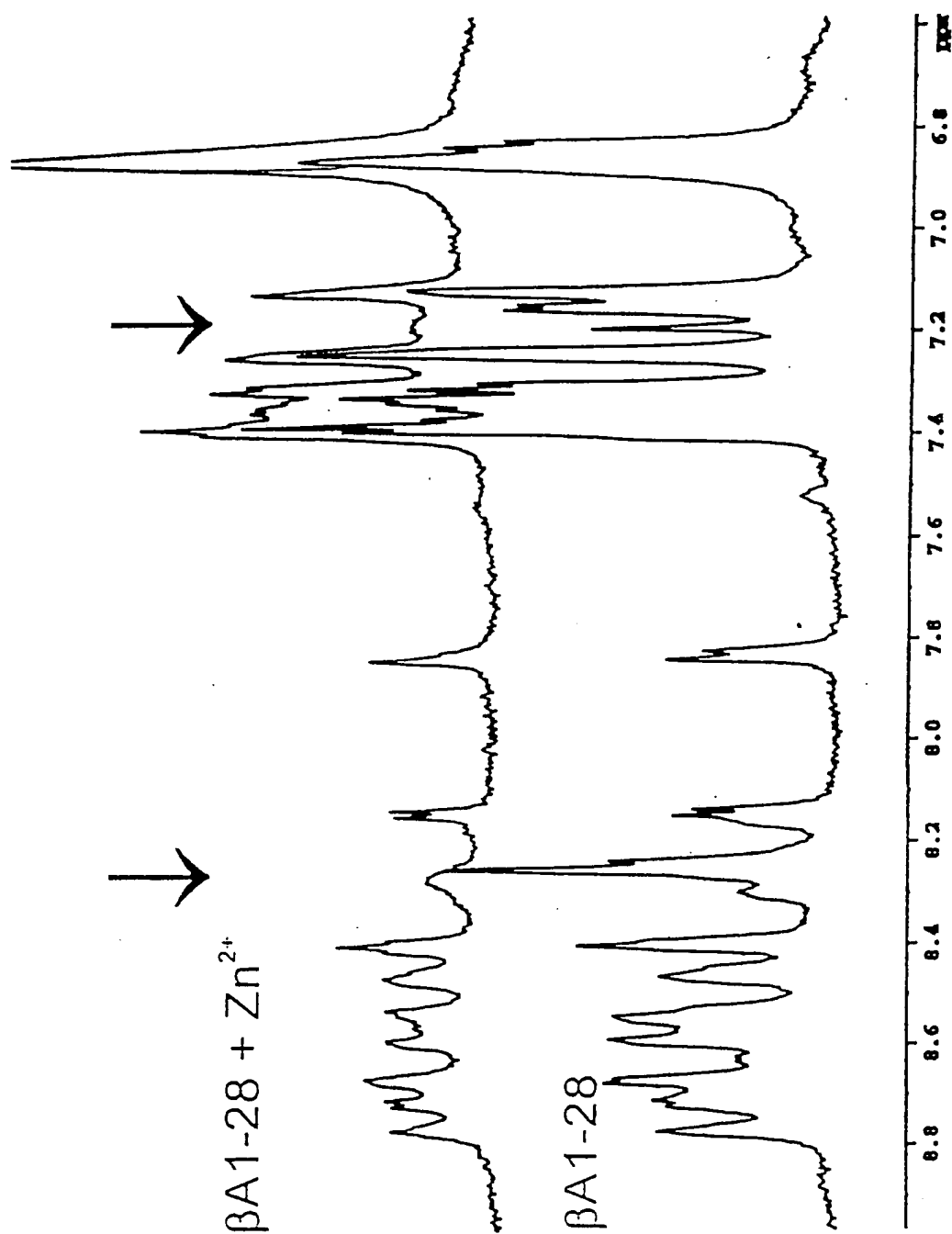
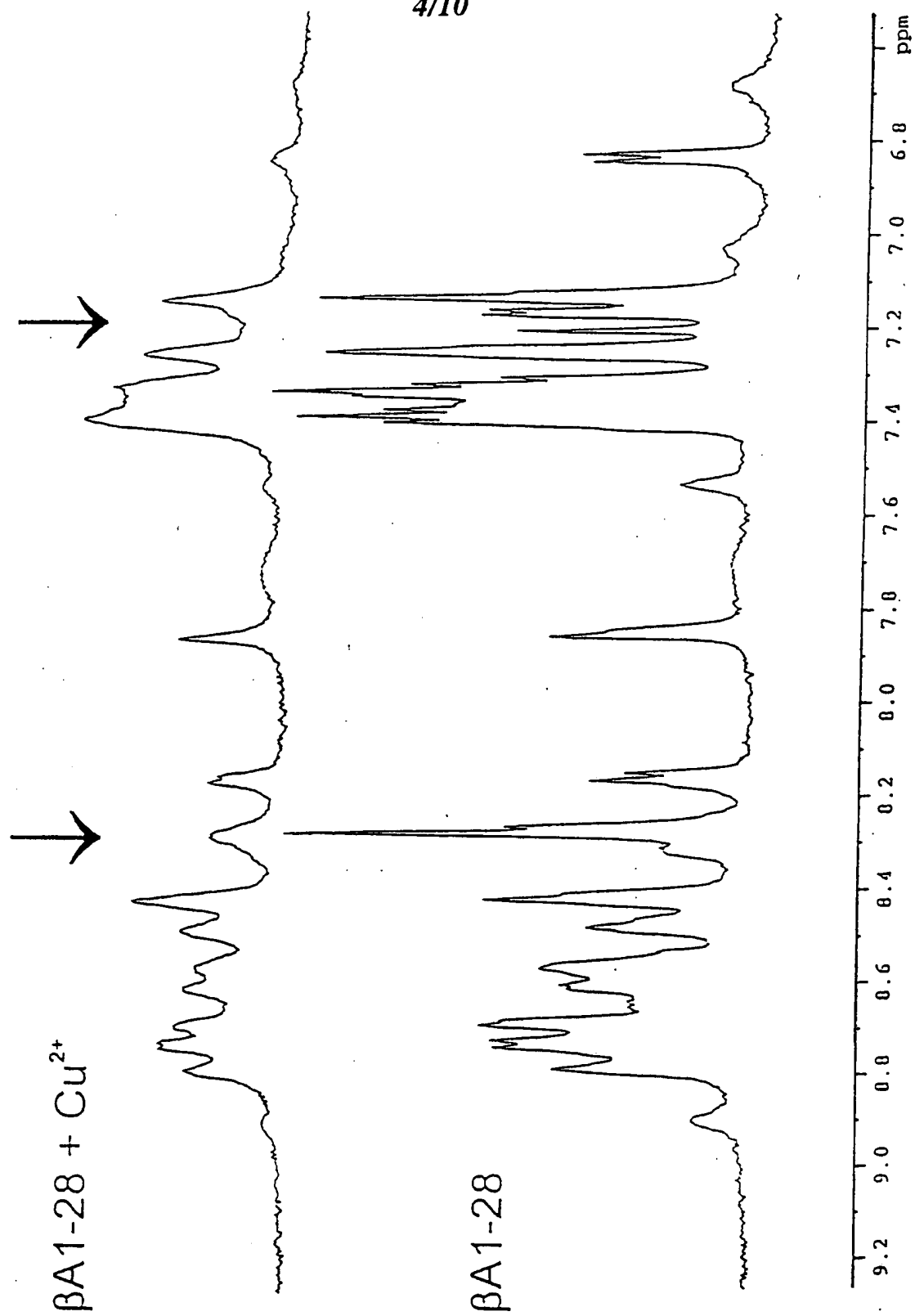
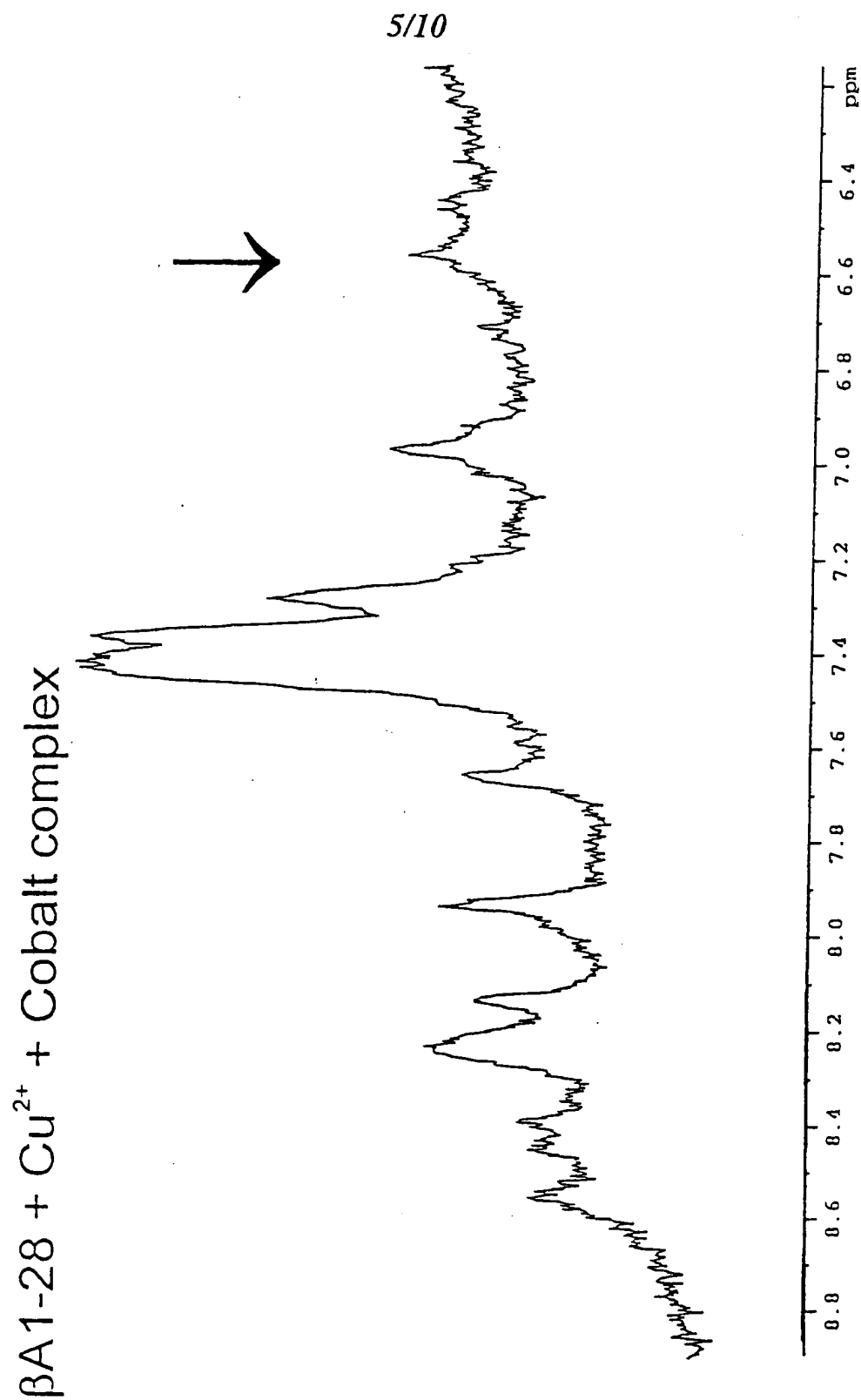
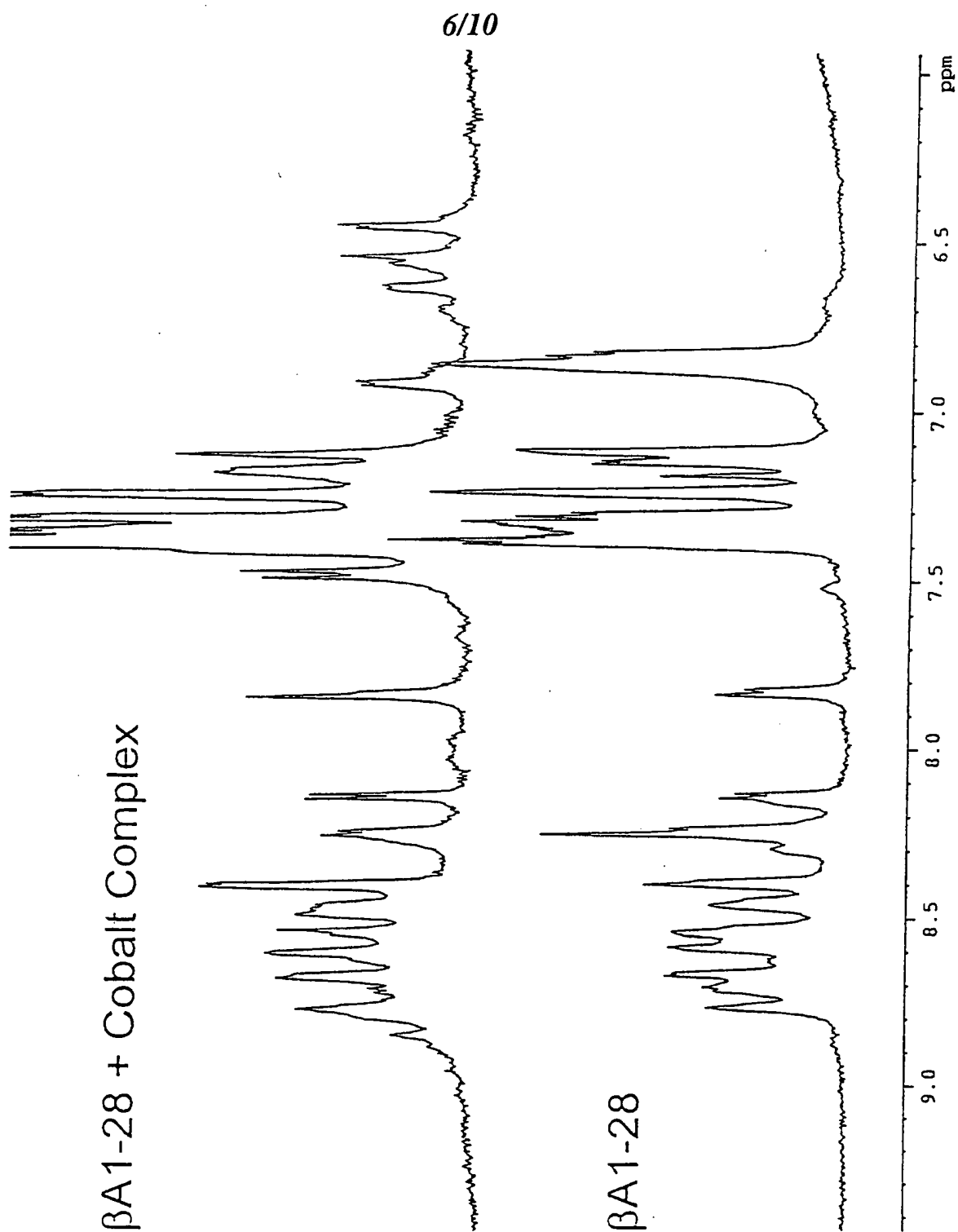


Figure 3

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**Figure 4**

*Figure 5*

**Figure 6**

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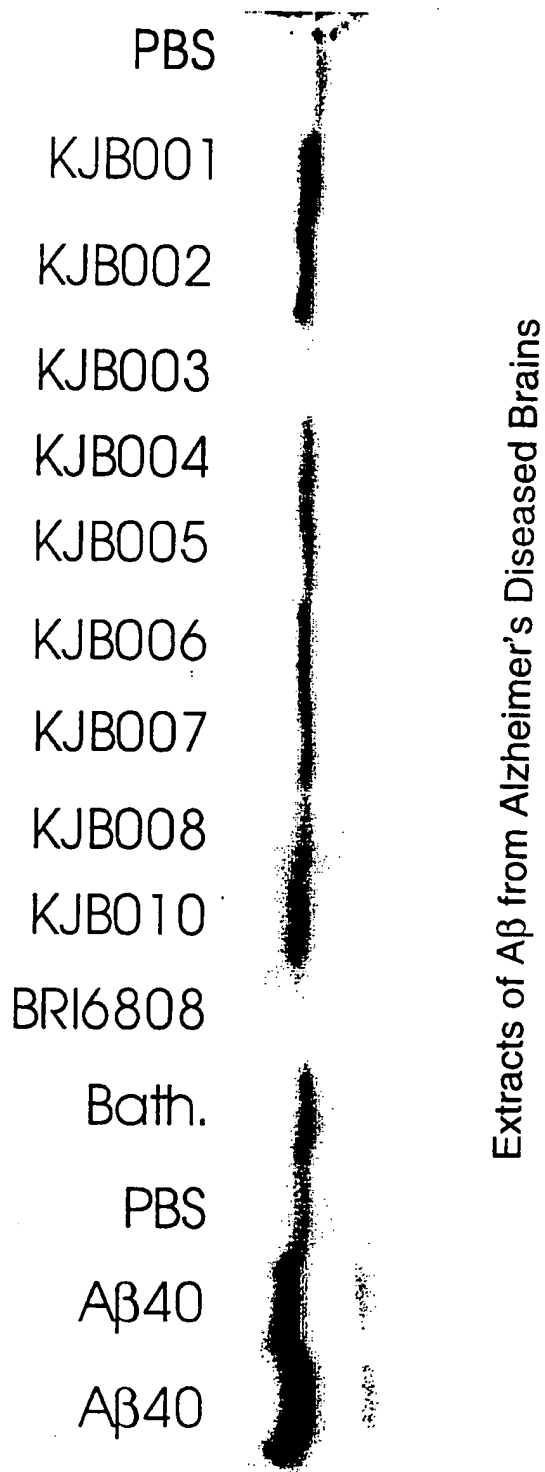


Figure 7

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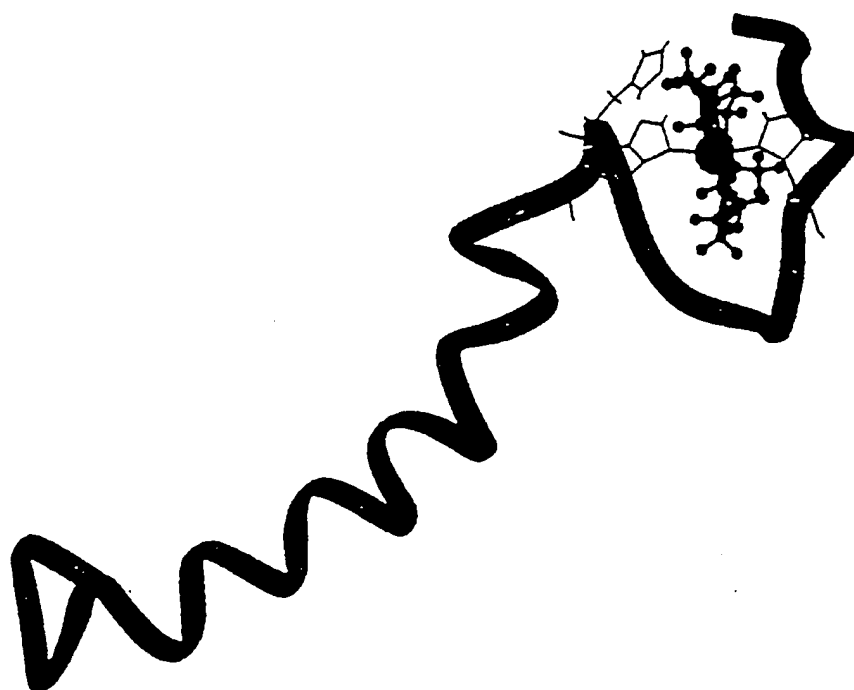
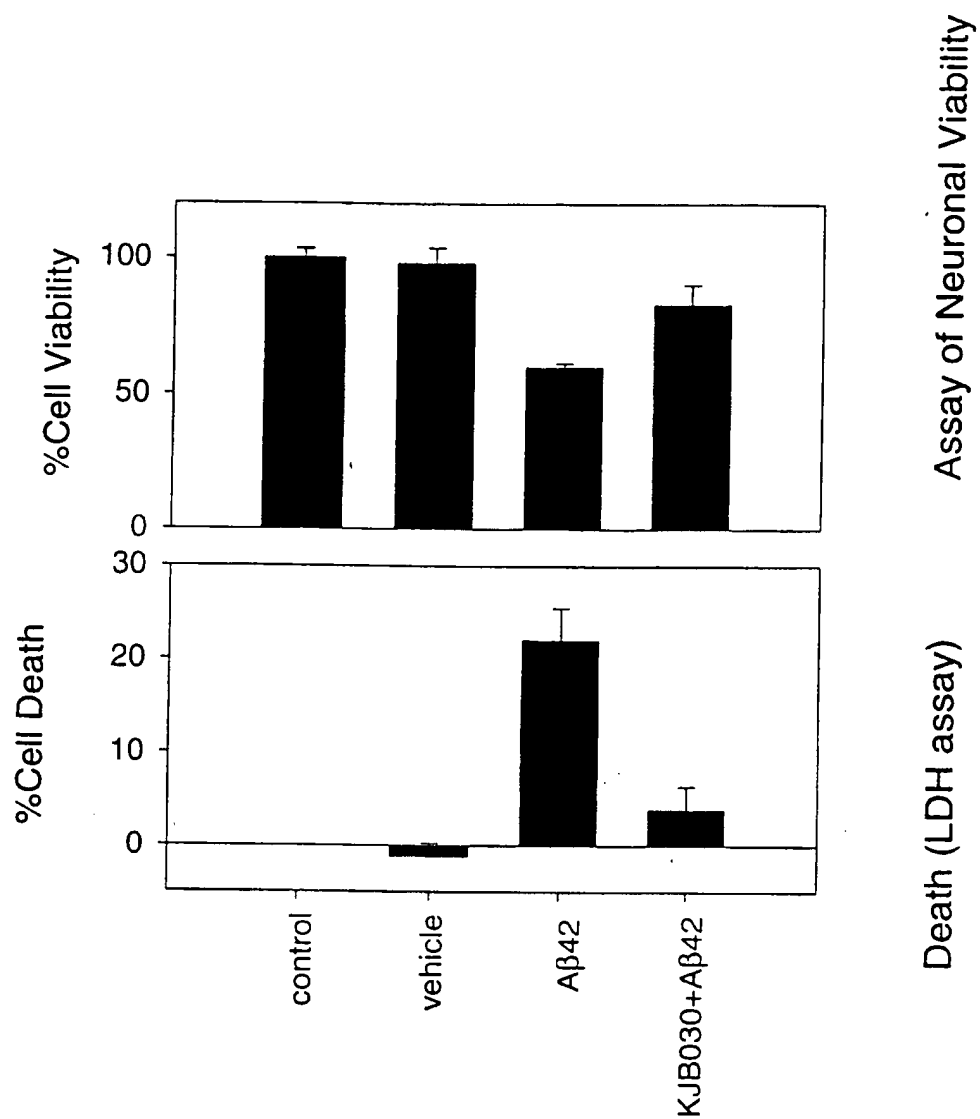
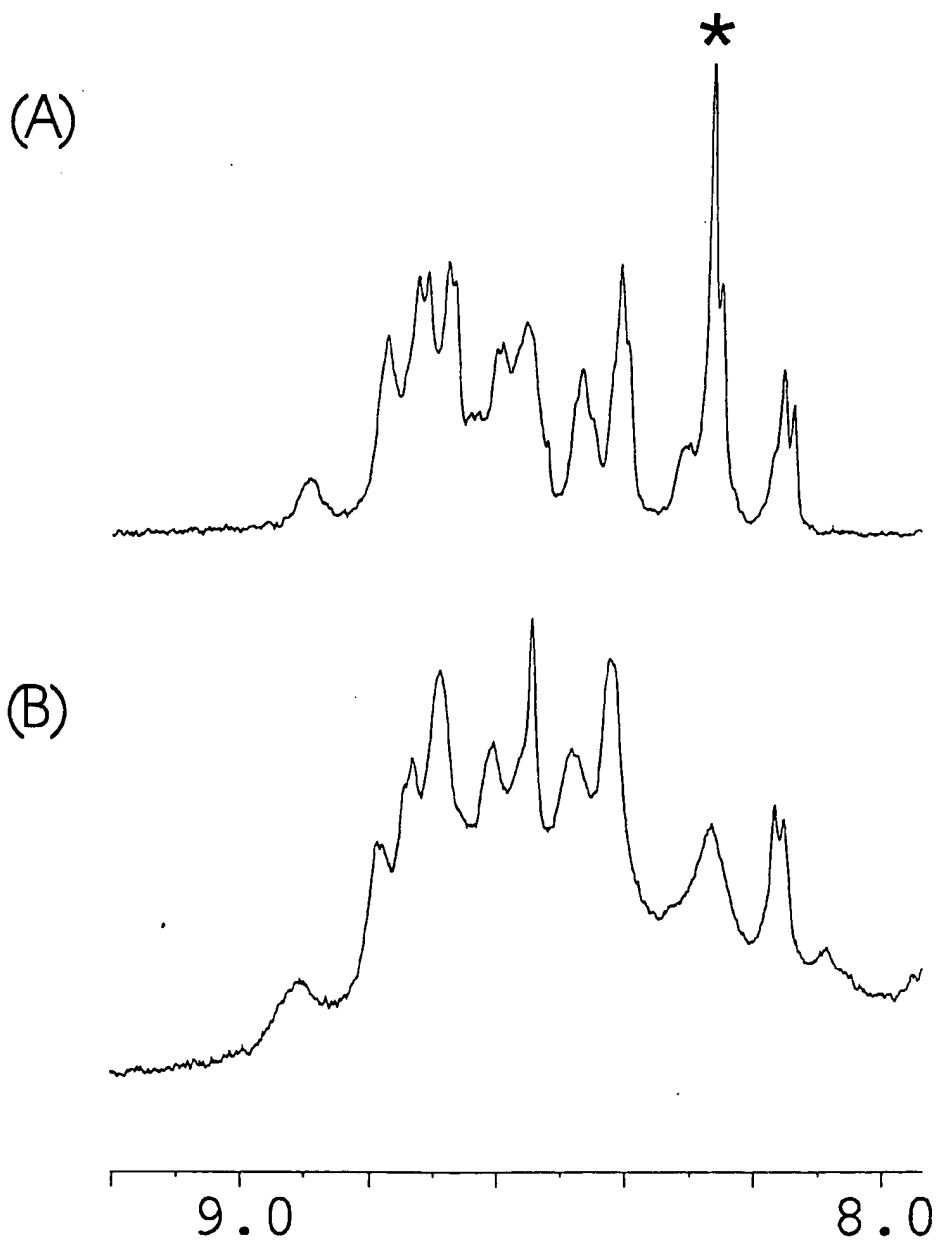


Figure 8

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*Figure 9*

10/10**Figure 10**

INTERNATIONAL SEARCH REPORT

International application No.
PCT/AU00/00886

A. CLASSIFICATION OF SUBJECT MATTER

Int. Cl. ⁷: C07D 487/22, 257/02, C07K 7/06, 14/47, 14/795, A61K 38/08, 38/41, A61P 25/28

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)
Database: STN, Files: CA, Medline, Biosis, WPIDS. Key words: beta amyloid, amyloid beta protein or peptide, abeta, N-term?, His 6, 13 or 14, inhib?, block?, destab?, compet? or bind?

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P,X	Biochemistry, volume 39, 2000, pages 7024-7031, T. Miura et al, "Metal binding modes of Alzheimer's amyloid β -peptide in insoluble aggregates and soluble complexes." Entire document.	1-42
X	Journal of Biological Chemistry, volume 273, no. 21, 1998, pages 12817-12826, C.S. Atwood et al, "Dramatic aggregation of Alzheimer A β by Cu(II) is induced by conditions representing physiological acidosis." Entire document and abstract.	1-42
X	Alzheimer's Research, volume 2, 1996, pages 189-194, L.J. Bartolotti et al, "A model for the tertiary structure of the β -amyloid peptide." See especially page 192, third paragraph.	1-42

☒ Further documents are listed in the continuation of Box C ☐ See patent family annex

* Special categories of cited documents:	
"A" document defining the general state of the art which is not considered to be of particular relevance	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"E" earlier application or patent but published on or after the international filing date	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"O" document referring to an oral disclosure, use, exhibition or other means	"&" document member of the same patent family
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search
31 August 2000

Date of mailing of the international search report
8 SEP 2000

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INTERNATIONAL SEARCH REPORT

International application No.

PCT/AU00/00886

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5958883 (Board of Regents of the University of Washington Office of Technology), 28 September 1999. Entire document, especially column 15 lines 60-66.	1-21
X	WO 95/12815 (The Research Foundation of State University of New York), 11 May 1995. Entire document, especially claim 2.	1-21
X	Chemical Abstracts 85:28019 & J. Chem. Soc., Dalton Transactions, 1976, no. 10, pages 858-862, P-K Chan et al, "Structural and mechanistic studies of coordination compounds. Part XIII. Syntheses and characterization of some dianiono(1,4,8,11-tetraazacyclotetradecane)manganese(III), - iron(III), and -nickel(III) salts. See abstract.	1-21
X	Journal of Molecular Biology, volume 285, January, 1999, pages 755-773, H. Shao et al, "Solution structures of micelle-bound amyloid β -(1-40) and β -(1-42) peptides of Alzheimer's disease." See page 767, left column, lines 54-60.	1-21
X	Journal of Neuroimmunology, volume 95, March, 1999, pages 136-142, D. Frenkel et al, "High affinity binding of monoclonal antibodies to the sequential epitope EFRH of β - amyloid peptide is essential for modulation of fibrillar aggregation." Entire document, especially page 141, second paragraph.	1-21
X	Journal of Biological Chemistry, volume 273, no. 13, 1998, pages 7185-7188, M. Pappolla et al, "Inhibition of Alzheimer β -fibrillogenesis by Melatonin." Entire document.	1-21
X	WO 98/44955 (Mindset Ltd.), 15 October 1998. See especially claim 1.	1-21
A	Biochemistry, volume 33, 1994, pages 7788-7796, J. Talafous et al, "Solution structure of residues 1-28 of the Amyloid β -peptide." Entire document, especially figure 3.	1-42
A	Journal of Biological Chemistry, volume 273, no. 45, 1998, pages 29719-29726, D. Giulian et al, "The HHQK domain of β -amyloid provides a structural basis for the immunopathology of Alzheimer's disease." Entire document.	1-42

INTERNATIONAL SEARCH REPORT

International application No.

PCT/AU00/00886

Box I Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☒ Claims Nos : 41
because they relate to subject matter not required to be searched by this Authority, namely:
This claim is to a method of treatment. Under rule 67.1 of the PCT this is excluded subject matter. However the search has been carried out based on the effects of the compound or pharmaceutical composition.
2. ☒ Claims Nos : 1-21
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
A full search was not possible on economic grounds. Claim 1 is inadequately defined. The documents cited are only a sample of possible compounds, including known compounds as described in the specification which inherently possess the properties as claimed in claim 1.
3. ☐ Claims Nos :
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a)

Box II Observations where unity of invention is lacking (Continuation of item 3 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.

INTERNATIONAL SEARCH REPORT
Information on patent family members

International application No.
PCT/AU00/00886

This Annex lists the known "A" publication level patent family members relating to the patent documents cited in the above-mentioned international search report. The Australian Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent Document Cited in Search Report				Patent Family Member			
WO	98/44955	AU	71034/98	CN	1254294	EP	994728
WO	95/12815	AU	81310/94	US	5744368		
							END OF ANNEX